

# «beta»-(4-Aminophenyl)propionic acid

<b>Other names:</b>	«beta»-(p-Aminophenyl)propionic acid p-Aminohydrocinnamic acid Benzenepropanoic acid, 4-amino- 3-(p-aminophenyl)propionic acid
<b>Inchi:</b>	InChI=1S/C9H11NO2/c10-8-4-1-7(2-5-8)3-6-9(11)12/h1-2,4-5H,3,6,10H2,(H,11,12)
<b>InchiKey:</b>	WXOHKMNWMKZMND-UHFFFAOYSA-N
<b>Formula:</b>	C9H11NO2
<b>SMILES:</b>	<chem>Nc1ccc(CCC(=O)O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	165.19
<b>CAS:</b>	2393-17-1

## Physical Properties

Property code	Value	Unit	Source
gf	-71.61	kJ/mol	Joback Method
hf	-235.05	kJ/mol	Joback Method
hfus	23.60	kJ/mol	Joback Method
hvap	72.63	kJ/mol	Joback Method
log10ws	-1.43		Crippen Method
logp	1.286		Crippen Method
mcvol	131.330	ml/mol	McGowan Method
pc	4140.93	kPa	Joback Method
tb	655.56	K	Joback Method
tc	867.32	K	Joback Method
tf	424.14	K	Joback Method
vc	0.485	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.06	J/molxK	655.56	Joback Method
cpg	343.03	J/molxK	690.85	Joback Method
cpg	352.35	J/molxK	726.15	Joback Method
cpg	361.05	J/molxK	761.44	Joback Method
cpg	369.15	J/molxK	796.73	Joback Method

cpg	376.69	J/mol×K	832.03	Joback Method
cpg	383.70	J/mol×K	867.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2393171&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2393171&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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