

# 1-(2,6-Dichlorophenyl)ethanol

<b>Inchi:</b>	InChI=1S/C8H8Cl2O/c1-5(11)8-6(9)3-2-4-7(8)10/h2-5,11H,1H3
<b>InchiKey:</b>	VUSOJMQVQGKPNN-UHFFFAOYSA-N
<b>Formula:</b>	C8H8Cl2O
<b>SMILES:</b>	CC(O)c1c(Cl)cccc1Cl
<b>Mol. weight [g/mol]:</b>	191.06

## Physical Properties

Property code	Value	Unit	Source
gf	-53.49	kJ/mol	Joback Method
hf	-183.85	kJ/mol	Joback Method
hfus	18.70	kJ/mol	Joback Method
hvap	62.06	kJ/mol	Joback Method
log10ws	-3.37		Crippen Method
logp	3.047		Crippen Method
mcvol	130.170	ml/mol	McGowan Method
pc	3633.35	kPa	Joback Method
tb	585.68	K	Joback Method
tc	798.96	K	Joback Method
tf	337.04	K	Joback Method
vc	0.486	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.61	J/mol×K	585.68	Joback Method
cpg	308.12	J/mol×K	763.41	Joback Method
cpg	301.24	J/mol×K	727.87	Joback Method
cpg	293.87	J/mol×K	692.32	Joback Method
cpg	285.99	J/mol×K	656.77	Joback Method
cpg	277.58	J/mol×K	621.23	Joback Method
cpg	314.53	J/mol×K	798.96	Joback Method
dvisc	0.0000959	Paxs	585.68	Joback Method
dvisc	0.0001434	Paxs	544.24	Joback Method
dvisc	0.0002291	Paxs	502.80	Joback Method

dvisc	0.0003981	Paxs	461.36	Joback Method
dvisc	0.0007718	Paxs	419.92	Joback Method
dvisc	0.0017296	Paxs	378.48	Joback Method
dvisc	0.0047267	Paxs	337.04	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=B6004857&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=B6004857&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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