

# 2-Hydroxy-2-methylbutyric acid

<b>Other names:</b>	Butanoic acid, 2-hydroxy-2-methyl- 2-Hydroxy-2-methylbutanoic acid
<b>Inchi:</b>	InChI=1S/C5H10O3/c1-3-5(2,8)4(6)7/h8H,3H2,1-2H3,(H,6,7)
<b>InchiKey:</b>	MBIQENSCDNJOIY-UHFFFAOYSA-N
<b>Formula:</b>	C5H10O3
<b>SMILES:</b>	CCC(C)(O)C(=O)O
<b>Mol. weight [g/mol]:</b>	118.13
<b>CAS:</b>	3739-30-8

## Physical Properties

Property code	Value	Unit	Source
gf	-408.50	kJ/mol	Joback Method
hf	-572.32	kJ/mol	Joback Method
hfus	11.07	kJ/mol	Joback Method
hvap	65.53	kJ/mol	Joback Method
log10ws	-0.39		Crippen Method
logp	0.232		Crippen Method
mvol	94.620	ml/mol	McGowan Method
pc	4876.56	kPa	Joback Method
rinpol	925.00		NIST Webbook
rinpol	925.00		NIST Webbook
tb	548.80	K	Joback Method
tc	724.01	K	Joback Method
tf	320.10	K	Joback Method
vc	0.348	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	228.03	J/molxK	548.80	Joback Method
cpg	235.28	J/molxK	578.00	Joback Method
cpg	242.14	J/molxK	607.20	Joback Method
cpg	248.61	J/molxK	636.41	Joback Method
cpg	254.72	J/molxK	665.61	Joback Method

cpg	260.49	J/mol×K	694.81	Joback Method
cpg	265.93	J/mol×K	724.01	Joback Method
dvisc	0.0296306	Paxs	320.10	Joback Method
dvisc	0.0061970	Paxs	358.22	Joback Method
dvisc	0.0017512	Paxs	396.33	Joback Method
dvisc	0.0006177	Paxs	434.45	Joback Method
dvisc	0.0002578	Paxs	472.57	Joback Method
dvisc	0.0001226	Paxs	510.68	Joback Method
dvisc	0.0000646	Paxs	548.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3739308&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3739308&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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