Benzamide, N,N-diheptyl-4-fluoro-

Inchi: InChI=1S/C21H34FNO/c1-3-5-7-9-11-17-23(18-12-10-8-6-4-2)21(24)19-13-15-20(22)16-
InchiKey: BUZIKMWJGOHCGQ-UHFFFAOYSA-N
Formula: C21H34FNO
SMILES: CCCCCCCN(CCCCCCC)C(=O)c1ccc(F)cc1
Mol. weight [g/mol]: 335.50

Physical Properties

Property code	Value	Unit	Source
gf	15.77	kJ/mol	Joback Method
hf	-492.87	kJ/mol	Joback Method
hfus	51.50	kJ/mol	Joback Method
hvap	73.25	kJ/mol	Joback Method
log10ws	-6.97		Crippen Method
logp	6.209		Crippen Method
mcvol	296.310	ml/mol	McGowan Method
pc	1187.42	kPa	Joback Method
rinpol	2341.00		NIST Webbook
tb	777.12	K	Joback Method
tc	963.23	K	Joback Method
tf	448.36	K	Joback Method
vc	1.145	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	904.17	J/mol×K	777.12	Joback Method
cpg	922.53	J/mol×K	808.14	Joback Method
cpg	939.87	J/mol×K	839.16	Joback Method
cpg	956.24	J/mol×K	870.17	Joback Method
cpg	971.68	J/mol×K	901.19	Joback Method
cpg	986.25	J/mol×K	932.21	Joback Method
cpg	1000.00	J/mol×K	963.23	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308308&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume