

3-Bromomethyl-6-bromo-7-sulfamoyl-3,4-dihydro-

Inchi:	InChI=1S/C8H9Br2N3O4S2/c9-3-8-12-5-1-4(10)6(18(11,14)15)2-7(5)19(16,17)13-8/h1-2
InchiKey:	UHLKDFTWNQPIKF-UHFFFAOYSA-N
Formula:	C8H9Br2N3O4S2
SMILES:	NS(=O)(=O)c1cc2c(cc1Br)NC(CBr)NS2(=O)=O
Mol. weight [g/mol]:	435.11
CAS:	3764-20-3

Physical Properties

Property code	Value	Unit	Source
gf	-511.18	kJ/mol	Joback Method
hf	-680.93	kJ/mol	Joback Method
hfus	62.62	kJ/mol	Joback Method
hvap	111.04	kJ/mol	Joback Method
log10ws	-3.15		Crippen Method
logp	0.521		Crippen Method
mcvol	210.080	ml/mol	McGowan Method
pc	9744.98	kPa	Joback Method
tb	811.63	K	Joback Method
tc	1062.41	K	Joback Method
tf	797.41	K	Joback Method
vc	0.795	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	507.63	J/mol×K	811.63	Joback Method
cpg	517.01	J/mol×K	853.43	Joback Method
cpg	525.21	J/mol×K	895.22	Joback Method
cpg	532.25	J/mol×K	937.02	Joback Method
cpg	538.14	J/mol×K	978.82	Joback Method
cpg	542.88	J/mol×K	1020.61	Joback Method
cpg	546.49	J/mol×K	1062.41	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C3764203&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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