

N-Methyl-3,4-methylenedioxyamphetamine

Other names:

MDMA
Ecstasy
Adam
1,3-Benzodioxole-5-ethanamine, N, «alpha»-dimethyl-
XTC
3,4-Methylenedioxyamphetamine
3,4-Methylenedioxyamphetamine
Methamphetamine, 3,4-methylenedioxy
Methylenedioxyamphetamine
(RS)-3,4-(Methylenedioxy)methamphetamine
3,4-Methylenedioxy-N, «alpha»-dimethyl-«beta»-phenylethylamine
DL-(3,4-Methylenedioxy)methamphetamine
Phenethylamine, N, «alpha»-dimethyl-3,4-methylenedioxy-
(. +/-)-(3,4-Methylenedioxy)methamphetamine
64057-70-1 (hydrochloride)
methylenedioxyamphetamine (MDMA,XTC)
Methylenedioxyamphetamine (MDMA)

Inchi:

InChI=1S/C11H15NO2/c1-8(12-2)5-9-3-4-10-11(6-9)14-7-13-10/h3-4,6,8,12H,5,7H2,1-2H

InchiKey:

SHXWCVYOXRDMCX-UHFFFAOYSA-N

Formula:

C11H15NO2

SMILES:

CNC(C)Cc1ccc2c(c1)OCO2

Mol. weight [g/mol]:

193.24

CAS:

42542-10-9

Physical Properties

Property code	Value	Unit	Source
gf	118.06	kJ/mol	Joback Method
hf	-179.45	kJ/mol	Joback Method
hfus	32.11	kJ/mol	Joback Method
hvap	58.97	kJ/mol	Joback Method
log10ws	-2.62		Crippen Method
logp	1.566		Crippen Method
mcvol	152.950	ml/mol	McGowan Method
pc	3062.55	kPa	Joback Method
rinpol	1551.10		NIST Webbook
rinpol	1553.00		NIST Webbook
rinpol	1553.00		NIST Webbook

rmpol	1560.00		NIST Webbook
rmpol	1560.00		NIST Webbook
rmpol	1554.00		NIST Webbook
rmpol	1551.10		NIST Webbook
rmpol	1585.00		NIST Webbook
tb	602.76	K	Joback Method
tc	824.22	K	Joback Method
tf	378.17	K	Joback Method
vc	0.573	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.82	J/mol×K	602.76	Joback Method
cpg	410.43	J/mol×K	639.67	Joback Method
cpg	424.06	J/mol×K	676.58	Joback Method
cpg	436.78	J/mol×K	713.49	Joback Method
cpg	448.65	J/mol×K	750.40	Joback Method
cpg	459.75	J/mol×K	787.31	Joback Method
cpg	470.13	J/mol×K	824.22	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C42542109&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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