

# Ethyl N,N-dimethyloxamate

<b>Other names:</b>	Acetic acid, (dimethylamino)oxo-, ethyl ester N,N-Dimethyl ethyl oxamate Oxamic acid, dimethyl-, ethyl ester ethyl (dimethylamino)oxoacetate
<b>Inchi:</b>	InChI=1S/C6H11NO3/c1-4-10-6(9)5(8)7(2)3/h4H2,1-3H3
<b>InchiKey:</b>	HMALWDVRMHVUAW-UHFFFAOYSA-N
<b>Formula:</b>	C6H11NO3
<b>SMILES:</b>	CCOC(=O)C(=O)N(C)C
<b>Mol. weight [g/mol]:</b>	145.16
<b>CAS:</b>	16703-52-9

## Physical Properties

Property code	Value	Unit	Source
gf	-252.42	kJ/mol	Joback Method
hf	-457.02	kJ/mol	Joback Method
hfus	18.70	kJ/mol	Joback Method
hvap	46.89	kJ/mol	Joback Method
ie	9.31	eV	NIST Webbook
log10ws	0.46		Crippen Method
logp	-0.362		Crippen Method
mcvol	114.390	ml/mol	McGowan Method
pc	3513.74	kPa	Joback Method
tb	479.28	K	Joback Method
tc	665.93	K	Joback Method
tf	311.94	K	Joback Method
vc	0.419	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.69	J/molxK	479.28	Joback Method
cpg	256.04	J/molxK	510.39	Joback Method
cpg	265.94	J/molxK	541.50	Joback Method
cpg	275.41	J/molxK	572.61	Joback Method

cpg	284.44	J/mol×K	603.72	Joback Method
cpg	293.05	J/mol×K	634.82	Joback Method
cpg	301.23	J/mol×K	665.93	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	402.50 ± 0.50	K	2.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16703529&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16703529&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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