

# 3-(3-Carboxy-propyl)-oxirane-2-carboxylic acid methyl ester

Inchi:	InChI=1S/C8H12O5/c1-12-8(11)7-5(13-7)3-2-4-6(9)10/h5,7H,2-4H2,1H3,(H,9,10)
InchiKey:	FSHLQCGOAFXLCO-UHFFFAOYSA-N
Formula:	C8H12O5
SMILES:	COC(=O)C1OC1CCCC(=O)O
Mol. weight [g/mol]:	188.18

## Physical Properties

Property code	Value	Unit	Source
gf	-516.26	kJ/mol	Joback Method
hf	-797.60	kJ/mol	Joback Method
hfus	32.13	kJ/mol	Joback Method
hvap	70.10	kJ/mol	Joback Method
log10ws	-0.34		Crippen Method
logp	0.182		Crippen Method
mcvol	133.470	ml/mol	McGowan Method
pc	3472.45	kPa	Joback Method
rinpol	1390.00		NIST Webbook
tb	633.80	K	Joback Method
tc	820.50	K	Joback Method
tf	403.10	K	Joback Method
vc	0.509	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	363.93	J/molxK	633.80	Joback Method
cpg	374.15	J/molxK	664.92	Joback Method
cpg	383.83	J/molxK	696.03	Joback Method
cpg	392.99	J/molxK	727.15	Joback Method
cpg	401.64	J/molxK	758.27	Joback Method
cpg	409.79	J/molxK	789.39	Joback Method
cpg	417.47	J/molxK	820.50	Joback Method
dvisc	0.0030986	Paxs	403.10	Joback Method
dvisc	0.0016701	Paxs	441.55	Joback Method

dvisc	0.0009938	Paxs	480.00	Joback Method
dvisc	0.0006387	Paxs	518.45	Joback Method
dvisc	0.0004364	Paxs	556.90	Joback Method
dvisc	0.0003132	Paxs	595.35	Joback Method
dvisc	0.0002340	Paxs	633.80	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R249251&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R249251&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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