

# Ethanone, 1-(1-cyclohexen-1-yl)-

<b>Other names:</b>	Ketone, 1-cyclohexen-1-yl methyl Methyl 1-cyclohexenyl ketone 1-Acetylcyclohexene 1-Cyclohexen-1-yl methyl ketone 1-Acetyl-1-cyclohexene 1-[1-cyclohexen-1-yl]-ethanone cyclohex-1-enylmethylketone
<b>Inchi:</b>	InChI=1S/C8H12O/c1-7(9)8-5-3-2-4-6-8/h5H,2-4,6H2,1H3
<b>InchiKey:</b>	LTYLUDGDHUEBGX-UHFFFAOYSA-N
<b>Formula:</b>	C8H12O
<b>SMILES:</b>	CC(=O)C1=CCCCC1
<b>Mol. weight [g/mol]:</b>	124.18
<b>CAS:</b>	932-66-1

## Physical Properties

Property code	Value	Unit	Source
gf	-59.95	kJ/mol	Joback Method
hf	-200.06	kJ/mol	Joback Method
hfus	9.67	kJ/mol	Joback Method
hvap	41.84	kJ/mol	Joback Method
log10ws	-2.20		Crippen Method
logp	2.076		Crippen Method
mcvol	109.990	ml/mol	McGowan Method
pc	3611.55	kPa	Joback Method
rinpol	1023.00		NIST Webbook
rinpol	931.00		NIST Webbook
rinpol	1023.00		NIST Webbook
tb	475.00 ± 7.00	K	NIST Webbook
tb	474.70	K	NIST Webbook
tc	681.88	K	Joback Method
tf	346.00	K	NIST Webbook
vc	0.409	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	223.02	J/molxK	464.67	Joback Method
cpg	237.48	J/molxK	500.87	Joback Method
cpg	251.13	J/molxK	537.07	Joback Method
cpg	263.98	J/molxK	573.27	Joback Method
cpg	276.08	J/molxK	609.48	Joback Method
cpg	287.44	J/molxK	645.68	Joback Method
cpg	298.08	J/molxK	681.88	Joback Method
dvisc	0.0044881	Paxs	254.75	Joback Method
dvisc	0.0021682	Paxs	289.74	Joback Method
dvisc	0.0012253	Paxs	324.72	Joback Method
dvisc	0.0007737	Paxs	359.71	Joback Method
dvisc	0.0005300	Paxs	394.70	Joback Method
dvisc	0.0003862	Paxs	429.68	Joback Method
dvisc	0.0002951	Paxs	464.67	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	474.50 ± 0.50	K	2.70	NIST Webbook

## Sources

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C932661&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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