

Phenol, 3-(1-methylethyl)-

Other names:	3-(1-Methylethyl)phenol 3-Isopropylphenol 3-isopropylhydroxybenzene Isopropylphenol, meta Phenol, m-isopropyl- m-Cumenol m-Isopropylphenol
Inchi:	InChI=1S/C9H12O/c1-7(2)8-4-3-5-9(10)6-8/h3-7,10H,1-2H3
InchiKey:	VLJSLTNSFSOYQR-UHFFFAOYSA-N
Formula:	C9H12O
SMILES:	CC(C)c1cccc(O)c1
Mol. weight [g/mol]:	136.19
CAS:	618-45-1

Physical Properties

Property code	Value	Unit	Source
chl	-4993.60	kJ/mol	NIST Webbook
gf	-19.75	kJ/mol	Joback Method
hf	-175.30 ± 2.40	kJ/mol	NIST Webbook
hf	-190.60	kJ/mol	NIST Webbook
hfl	-260.00	kJ/mol	NIST Webbook
hfus	15.37	kJ/mol	Joback Method
hvap	69.40	kJ/mol	NIST Webbook
hvap	73.01	kJ/mol	NIST Webbook
log10ws	-2.20		Crippen Method
logp	2.516		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3935.71	kPa	Joback Method
rinpol	1227.30		NIST Webbook
rinpol	1207.00		NIST Webbook
rinpol	1228.00		NIST Webbook
rinpol	1229.20		NIST Webbook
rinpol	1227.90		NIST Webbook
rinpol	1207.00		NIST Webbook
tb	512.18	K	Joback Method
tc	739.56	K	Joback Method
tf	299.15 ± 2.00	K	NIST Webbook

tf	299.15 ± 2.00	K	NIST Webbook
tf	299.15 ± 2.00	K	NIST Webbook
tf	299.15 ± 2.00	K	NIST Webbook
tf	298.85 ± 0.20	K	NIST Webbook
tf	299.15 ± 2.00	K	NIST Webbook
vc	0.392	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	267.53	J/mol×K	512.18	Joback Method
cpg	280.87	J/mol×K	550.08	Joback Method
cpg	293.23	J/mol×K	587.97	Joback Method
cpg	304.71	J/mol×K	625.87	Joback Method
cpg	315.38	J/mol×K	663.76	Joback Method
cpg	325.34	J/mol×K	701.66	Joback Method
cpg	334.65	J/mol×K	739.56	Joback Method
dvisc	0.0072081	Paxs	314.33	Joback Method
dvisc	0.0023278	Paxs	347.31	Joback Method
dvisc	0.0009146	Paxs	380.28	Joback Method
dvisc	0.0004171	Paxs	413.25	Joback Method
dvisc	0.0002136	Paxs	446.23	Joback Method
dvisc	0.0001200	Paxs	479.21	Joback Method
dvisc	0.0000726	Paxs	512.18	Joback Method
hvapt	64.30	kJ/mol	437.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54407e+01
Coeff. B	-4.55275e+03
Coeff. C	-8.04700e+01
Temperature range (K), min.	380.92
Temperature range (K), max.	529.94

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C618451&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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