

# 2-chloropentyl trichloroacetate

<b>Other names:</b>	1-Pentanol, 2-chloro, trichloroacetate
<b>Inchi:</b>	InChI=1S/C7H10Cl4O2/c1-2-3-5(8)4-13-6(12)7(9,10)11/h5H,2-4H2,1H3
<b>InchiKey:</b>	KLZWAXURZZRALT-UHFFFAOYSA-N
<b>Formula:</b>	C7H10Cl4O2
<b>SMILES:</b>	CCCC(Cl)COC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	267.96

## Physical Properties

Property code	Value	Unit	Source
gf	-273.18	kJ/mol	Joback Method
hf	-509.60	kJ/mol	Joback Method
hfus	22.52	kJ/mol	Joback Method
hvap	56.19	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.307		Crippen Method
mcvol	165.890	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
rinpol	1387.00		NIST Webbook
rinpol	1348.00		NIST Webbook
rinpol	1398.00		NIST Webbook
rinpol	1393.00		NIST Webbook
ripol	1920.00		NIST Webbook
ripol	1935.00		NIST Webbook
ripol	1908.00		NIST Webbook
ripol	1905.00		NIST Webbook
tb	581.90	K	Joback Method
tc	795.32	K	Joback Method
tf	347.91	K	Joback Method
vc	0.630	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.31	J/mol×K	581.90	Joback Method

cpg	385.77	J/molxK	759.75	Joback Method
cpg	378.32	J/molxK	724.18	Joback Method
cpg	370.28	J/molxK	688.61	Joback Method
cpg	361.62	J/molxK	653.04	Joback Method
cpg	352.30	J/molxK	617.47	Joback Method
cpg	392.66	J/molxK	795.32	Joback Method
dvisc	0.0002097	Paxs	581.90	Joback Method
dvisc	0.0002785	Paxs	542.90	Joback Method
dvisc	0.0003867	Paxs	503.90	Joback Method
dvisc	0.0005673	Paxs	464.90	Joback Method
dvisc	0.0008926	Paxs	425.91	Joback Method
dvisc	0.0015388	Paxs	386.91	Joback Method
dvisc	0.0029975	Paxs	347.91	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R112397&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R112397&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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