

# Dimethylmalonic acid, 4-acetylphenyl ethyl ester

Inchi:	InChI=1S/C15H18O5/c1-5-19-13(17)15(3,4)14(18)20-12-8-6-11(7-9-12)10(2)16/h6-9H,5H
InchiKey:	MFTKKTMKNNWZRH-UHFFFAOYSA-N
Formula:	C15H18O5
SMILES:	CCOC(=O)C(C)(C)C(=O)Oc1ccc(C(C)=O)cc1
Mol. weight [g/mol]:	278.30

## Physical Properties

Property code	Value	Unit	Source
gf	-415.72	kJ/mol	Joback Method
hf	-738.80	kJ/mol	Joback Method
hfus	28.02	kJ/mol	Joback Method
hvap	75.68	kJ/mol	Joback Method
log10ws	-3.16		Crippen Method
logp	2.384		Crippen Method
mvol	214.900	ml/mol	McGowan Method
pc	2127.56	kPa	Joback Method
rinpol	1984.00		NIST Webbook
rinpol	1984.00		NIST Webbook
tb	777.48	K	Joback Method
tc	996.60	K	Joback Method
tf	494.42	K	Joback Method
vc	0.810	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	611.30	J/molxK	777.48	Joback Method
cpg	624.71	J/molxK	814.00	Joback Method
cpg	637.05	J/molxK	850.52	Joback Method
cpg	648.37	J/molxK	887.04	Joback Method
cpg	658.68	J/molxK	923.56	Joback Method
cpg	668.02	J/molxK	960.08	Joback Method
cpg	676.42	J/molxK	996.60	Joback Method
dvisc	0.0007284	Paxs	494.42	Joback Method

dvisc	0.0004348	Paxs	541.60	Joback Method
dvisc	0.0002820	Paxs	588.77	Joback Method
dvisc	0.0001950	Paxs	635.95	Joback Method
dvisc	0.0001419	Paxs	683.13	Joback Method
dvisc	0.0001075	Paxs	730.30	Joback Method
dvisc	0.0000843	Paxs	777.48	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363697&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363697&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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