

# 2,4-Cycloheptadien-1-one, 2,6,6-trimethyl-

<b>Other names:</b>	Eucarvone 2,6,6-Trimethyl-2,4-cycloheptadien-1-one
<b>Inchi:</b>	InChI=1S/C10H14O/c1-8-5-4-6-10(2,3)7-9(8)11/h4-6H,7H2,1-3H3
<b>InchiKey:</b>	QNGQIURXCUHNAT-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CC1=CC=CC(C)(C)CC1=O
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	503-93-5

## Physical Properties

Property code	Value	Unit	Source
gf	-32.12	kJ/mol	Joback Method
hf	-219.94	kJ/mol	Joback Method
hfus	6.66	kJ/mol	Joback Method
hvap	42.80	kJ/mol	Joback Method
ie	9.62	eV	NIST Webbook
log10ws	-2.65		Crippen Method
logp	2.488		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3049.04	kPa	Joback Method
rinpol	1222.70		NIST Webbook
rinpol	1233.00		NIST Webbook
rinpol	1243.00		NIST Webbook
rinpol	1233.00		NIST Webbook
rinpol	1222.70		NIST Webbook
rinpol	1245.00		NIST Webbook
rinpol	1248.00		NIST Webbook
tb	523.38	K	Joback Method
tc	760.57	K	Joback Method
tf	312.48	K	Joback Method
vc	0.497	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.42	J/mol×K	523.38	Joback Method
cpg	320.39	J/mol×K	562.91	Joback Method
cpg	336.37	J/mol×K	602.44	Joback Method
cpg	351.45	J/mol×K	641.98	Joback Method
cpg	365.72	J/mol×K	681.51	Joback Method
cpg	379.28	J/mol×K	721.04	Joback Method
cpg	392.23	J/mol×K	760.57	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C503935&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C503935&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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