

# Epoxypseudoisoeugenyl 2-methylbutyrate

<b>Inchi:</b>	InChI=1S/C14H18O4/c1-8(2)14(15)18-11-6-5-10(7-12(11)16-4)13-9(3)17-13/h5-9,13H,1-
<b>InchiKey:</b>	MCVNHWFLEYXICF-UHFFFAOYSA-N
<b>Formula:</b>	C14H18O4
<b>SMILES:</b>	COc1cc(C2OC2C)ccc1OC(=O)C(C)C
<b>Mol. weight [g/mol]:</b>	250.29

## Physical Properties

Property code	Value	Unit	Source
gf	-214.29	kJ/mol	Joback Method
hf	-580.54	kJ/mol	Joback Method
hfus	32.92	kJ/mol	Joback Method
hvap	65.65	kJ/mol	Joback Method
log10ws	-3.27		Crippen Method
logp	2.716		Crippen Method
mcvol	192.680	ml/mol	McGowan Method
pc	2191.78	kPa	Joback Method
rinpol	1844.00		NIST Webbook
rinpol	1844.00		NIST Webbook
rinpol	1844.00		NIST Webbook
rinpol	1850.00		NIST Webbook
rinpol	1850.00		NIST Webbook
ripol	2698.00		NIST Webbook
ripol	2698.00		NIST Webbook
ripol	2698.00		NIST Webbook
tb	683.65	K	Joback Method
tc	898.45	K	Joback Method
tf	418.66	K	Joback Method
vc	0.725	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	540.12	J/mol×K	683.65	Joback Method
cpg	610.82	J/mol×K	862.65	Joback Method

cpg	598.59	J/molxK	826.85	Joback Method
cpg	585.42	J/molxK	791.05	Joback Method
cpg	571.30	J/molxK	755.25	Joback Method
cpg	556.21	J/molxK	719.45	Joback Method
cpg	622.14	J/molxK	898.45	Joback Method
dvisc	0.0003215	Paxs	683.65	Joback Method
dvisc	0.0003760	Paxs	639.49	Joback Method
dvisc	0.0004502	Paxs	595.32	Joback Method
dvisc	0.0005547	Paxs	551.15	Joback Method
dvisc	0.0007088	Paxs	506.99	Joback Method
dvisc	0.0009491	Paxs	462.83	Joback Method
dvisc	0.0013517	Paxs	418.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R240041&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R240041&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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