

# Glutaric acid, di(2-chlorophenyl) ester

<b>Inchi:</b>	InChI=1S/C17H14Cl2O4/c18-12-6-1-3-8-14(12)22-16(20)10-5-11-17(21)23-15-9-4-2-7-13
<b>InchiKey:</b>	MBYBZTSTRXWQLC-UHFFFAOYSA-N
<b>Formula:</b>	C17H14Cl2O4
<b>SMILES:</b>	O=C(CCCC(=O)Oc1ccccc1Cl)Oc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	353.20

## Physical Properties

Property code	Value	Unit	Source
gf	-193.88	kJ/mol	Joback Method
hf	-465.17	kJ/mol	Joback Method
hfus	41.06	kJ/mol	Joback Method
hvap	86.39	kJ/mol	Joback Method
log10ws	-5.54		Crippen Method
logp	4.675		Crippen Method
mcvol	242.230	ml/mol	McGowan Method
pc	2066.12	kPa	Joback Method
rinqol	2717.00		NIST Webbook
tb	879.12	K	Joback Method
tc	1116.93	K	Joback Method
tf	563.39	K	Joback Method
vc	0.917	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.24	J/molxK	879.12	Joback Method
cpg	698.12	J/molxK	1077.30	Joback Method
cpg	691.83	J/molxK	1037.66	Joback Method
cpg	684.43	J/molxK	998.03	Joback Method
cpg	675.88	J/molxK	958.39	Joback Method
cpg	666.17	J/molxK	918.76	Joback Method
cpg	703.34	J/molxK	1116.93	Joback Method
dvisc	0.0000651	Paxs	879.12	Joback Method
dvisc	0.0000807	Paxs	826.50	Joback Method

dvisc	0.0001029	Paxs	773.88	Joback Method
dvisc	0.0001359	Paxs	721.25	Joback Method
dvisc	0.0001876	Paxs	668.63	Joback Method
dvisc	0.0002737	Paxs	616.01	Joback Method
dvisc	0.0004284	Paxs	563.39	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358608&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358608&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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