

# Benzene, 1-(chloromethyl)-2-fluoro-

<b>Other names:</b>	o-Fluorobenzyl chloride 2-Fluorobenzyl chloride «alpha»-Chloro-o-fluorotoluene «alpha»-Chloro-2-fluorotoluene Toluene, «alpha»-chloro-o-fluoro- 1-(Chloromethyl)-2-fluorobenzene
<b>Inchi:</b>	InChI=1S/C7H6ClF/c8-5-6-3-1-2-4-7(6)9/h1-4H,5H2
<b>InchiKey:</b>	MOBRMRJUKNQBMU-UHFFFAOYSA-N
<b>Formula:</b>	C7H6ClF
<b>SMILES:</b>	Fc1ccccc1CCl
<b>Mol. weight [g/mol]:</b>	144.57
<b>CAS:</b>	345-35-7

## Physical Properties

Property code	Value	Unit	Source
gf	-95.90	kJ/mol	Joback Method
hf	-174.60	kJ/mol	Joback Method
hfus	14.81	kJ/mol	Joback Method
hvap	37.68	kJ/mol	Joback Method
log10ws	-2.84		Crippen Method
logp	2.564		Crippen Method
mvol	99.740	ml/mol	McGowan Method
pc	3646.53	kPa	Joback Method
rinpol	983.70		NIST Webbook
rinpol	983.70		NIST Webbook
tb	427.92	K	Joback Method
tc	636.83	K	Joback Method
tf	238.10	K	Joback Method
vc	0.387	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	172.12	J/mol×K	427.92	Joback Method

cpg	181.95	J/mol×K	462.74	Joback Method
cpg	191.20	J/mol×K	497.56	Joback Method
cpg	199.89	J/mol×K	532.38	Joback Method
cpg	208.06	J/mol×K	567.20	Joback Method
cpg	215.72	J/mol×K	602.01	Joback Method
cpg	222.90	J/mol×K	636.83	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	359.00	K	5.30	NIST Webbook
tbrp	359.20	K	5.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C345357&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C345357&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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