

# 1,2-Dithiane

<b>Other names:</b>	o-Dithiane Tetramethylene disulfide 1,2-Dithiacyclohexane
<b>Inchi:</b>	InChI=1S/C4H8S2/c1-2-4-6-5-3-1/h1-4H2
<b>InchiKey:</b>	CXWGKAYMVASWDQ-UHFFFAOYSA-N
<b>Formula:</b>	C4H8S2
<b>SMILES:</b>	C1CCSSC1
<b>Mol. weight [g/mol]:</b>	120.24
<b>CAS:</b>	505-20-4

## Physical Properties

Property code	Value	Unit	Source
gf	94.68	kJ/mol	Joback Method
hf	39.29	kJ/mol	Joback Method
hfus	4.19	kJ/mol	Joback Method
hvap	36.86	kJ/mol	Joback Method
ie	8.36	eV	NIST Webbook
ie	8.10	eV	NIST Webbook
log10ws	-2.15		Crippen Method
logp	2.162		Crippen Method
mcvol	89.060	ml/mol	McGowan Method
pc	5281.57	kPa	Joback Method
rinpol	1088.00		NIST Webbook
rinpol	1088.00		NIST Webbook
rinpol	1086.00		NIST Webbook
ripol	1475.00		NIST Webbook
ripol	1489.00		NIST Webbook
ripol	1489.00		NIST Webbook
tb	410.80	K	Joback Method
tc	657.97	K	Joback Method
tf	313.36	K	Joback Method
vc	0.285	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	146.23	J/mol×K	410.80	Joback Method
cpg	158.39	J/mol×K	451.99	Joback Method
cpg	169.74	J/mol×K	493.19	Joback Method
cpg	180.29	J/mol×K	534.38	Joback Method
cpg	190.10	J/mol×K	575.58	Joback Method
cpg	199.20	J/mol×K	616.77	Joback Method
cpg	207.62	J/mol×K	657.97	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	333.20	K	0.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C505204&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C505204&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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