

# 1,3-Dioxolane, 2-ethyl-2-methyl-

<b>Other names:</b>	2-Butanone, cyclic ethylene acetal 2-Butanone, cyclic 1,2-ethanediyl acetal 2-Ethyl-2-methyldioxolane 2-Methyl-2-ethyldioxolane 2-Ethyl-2-methyl-1,3-dioxolane 2-Methyl-2-ethyl-1,3-dioxolane
<b>Inchi:</b>	InChI=1S/C6H12O2/c1-3-6(2)7-4-5-8-6/h3-5H2,1-2H3
<b>InchiKey:</b>	UPZFLZYXYGBAPL-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O2
<b>SMILES:</b>	CCC1(C)OCCO1
<b>Mol. weight [g/mol]:</b>	116.16
<b>CAS:</b>	126-39-6

## Physical Properties

Property code	Value	Unit	Source
gf	-141.54	kJ/mol	Joback Method
hf	-355.45	kJ/mol	Joback Method
hfus	14.89	kJ/mol	Joback Method
hvap	44.80 ± 0.30	kJ/mol	NIST Webbook
log10ws	-1.01		Crippen Method
logp	1.159		Crippen Method
mcvol	96.280	ml/mol	McGowan Method
pc	3970.51	kPa	Joback Method
rinpol	766.00		NIST Webbook
rinpol	766.00		NIST Webbook
rinpol	768.00		NIST Webbook
rinpol	779.00		NIST Webbook
tb	391.20	K	NIST Webbook
tb	391.05	K	NIST Webbook
tb	389.70	K	NIST Webbook
tc	612.10	K	Joback Method
tf	191.19 ± 0.60	K	NIST Webbook
tf	191.15	K	NIST Webbook
vc	0.352	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.72	J/mol×K	577.77	Joback Method
cpg	195.53	J/mol×K	406.10	Joback Method
cpg	209.42	J/mol×K	440.43	Joback Method
cpg	222.29	J/mol×K	474.77	Joback Method
cpg	234.24	J/mol×K	509.10	Joback Method
cpg	245.35	J/mol×K	543.43	Joback Method
cpg	265.44	J/mol×K	612.10	Joback Method
hvapt	43.10 ± 0.30	kJ/mol	293.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C126396&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C126396&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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