

# Butanoic acid, pentyl ester

<b>Other names:</b>	1-Pentyl butyrate AMYL N-BUTYRATE Amyl butanoate Amyl butyrate Butyric acid, pentyl ester PENTYL BUTANOATE Pentyl butyrate UN 2620 n-Amyl butyrate n-Amyl n-butyrate n-Pentyl butanoate n-Pentyl butyrate n-Pentyl n-butyrate
<b>Inchi:</b>	InChI=1S/C9H18O2/c1-3-5-6-8-11-9(10)7-4-2/h3-8H2,1-2H3
<b>InchiKey:</b>	CFNJLPHOBMVMNS-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O2
<b>SMILES:</b>	CCCCCOC(=O)CCC
<b>Mol. weight [g/mol]:</b>	158.24
<b>CAS:</b>	540-18-1

## Physical Properties

Property code	Value	Unit	Source
gf	-209.02	kJ/mol	Joback Method
hf	-473.89	kJ/mol	Joback Method
hfus	21.85	kJ/mol	Joback Method
hvap	44.78	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	2.520		Crippen Method
mcvol	145.110	ml/mol	McGowan Method
pc	2412.37	kPa	Joback Method
rinpol	1055.00		NIST Webbook
rinpol	1079.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1091.00		NIST Webbook
rinpol	1062.00		NIST Webbook
rinpol	1044.00		NIST Webbook

rinpol	1052.00		NIST Webbook
rinpol	1077.00		NIST Webbook
rinpol	1088.00		NIST Webbook
rinpol	1084.00		NIST Webbook
rinpol	1083.00		NIST Webbook
rinpol	1076.00		NIST Webbook
rinpol	1037.00		NIST Webbook
rinpol	1076.00		NIST Webbook
rinpol	1062.00		NIST Webbook
rinpol	1076.00		NIST Webbook
rinpol	1075.00		NIST Webbook
rinpol	1082.00		NIST Webbook
rinpol	1075.00		NIST Webbook
rinpol	1039.00		NIST Webbook
rinpol	1039.00		NIST Webbook
rinpol	1080.00		NIST Webbook
rinpol	1081.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1080.00		NIST Webbook
rinpol	1062.00		NIST Webbook
rinpol	1096.00		NIST Webbook
rinpol	1098.00		NIST Webbook
rinpol	1094.00		NIST Webbook
rinpol	1062.00		NIST Webbook
rinpol	1092.00		NIST Webbook
rinpol	1095.00		NIST Webbook
rinpol	1091.00		NIST Webbook
rinpol	1093.00		NIST Webbook
rinpol	1059.00		NIST Webbook
rinpol	1054.00		NIST Webbook
ripol	1304.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1320.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1320.00		NIST Webbook
ripol	1326.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1319.00		NIST Webbook
ripol	1340.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1321.00		NIST Webbook
ripol	1304.00		NIST Webbook
tb	451.75 ± 1.00	K	NIST Webbook

tb	449.00 ± 2.00	K	NIST Webbook
tb	459.50 ± 0.50	K	NIST Webbook
tc	655.29	K	Joback Method
tf	263.35	K	Joback Method
tt	200.48 ± 0.02	K	NIST Webbook
vc	0.564	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	322.89	J/mol×K	481.61	Joback Method
cpg	336.15	J/mol×K	510.56	Joback Method
cpg	348.92	J/mol×K	539.50	Joback Method
cpg	361.21	J/mol×K	568.45	Joback Method
cpg	373.04	J/mol×K	597.40	Joback Method
cpg	384.40	J/mol×K	626.35	Joback Method
cpg	395.29	J/mol×K	655.29	Joback Method
dvisc	0.0033103	Paxs	263.35	Joback Method
dvisc	0.0016384	Paxs	299.73	Joback Method
dvisc	0.0009442	Paxs	336.10	Joback Method
dvisc	0.0006060	Paxs	372.48	Joback Method
dvisc	0.0004209	Paxs	408.86	Joback Method
dvisc	0.0003103	Paxs	445.23	Joback Method
dvisc	0.0002395	Paxs	481.61	Joback Method
hvapt	53.60	kJ/mol	298.15	Vapor pressures and vaporization enthalpies of a series of esters used in flavors by correlation gas chromatography

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.56224e+01
Coeff. B	-4.21714e+03
Coeff. C	-6.85160e+01

Temperature range (K), min.	343.52
Temperature range (K), max.	477.51

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	4.34814e+01
Coeff. B	-6.74039e+03
Coeff. C	-3.98394e+00
Coeff. D	1.98902e-06
Temperature range (K), min.	365.15
Temperature range (K), max.	477.15

## Sources

<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1114.mol">https://www.thermo.com/files/research/kdb/mol/mol1114.mol</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="https://www.chemed.com/doc/models/crippen_log10ws">https://www.chemed.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C540181&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C540181&amp;Units=SI</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1114">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1114</a>
<b>Vapor pressures and vaporization enthalpies of a series of esters used in flavor by correlation gas chromatography:</b>	<a href="https://www.doi.org/10.1016/j.jct.2015.02.015">https://www.doi.org/10.1016/j.jct.2015.02.015</a>
<b>McGowan Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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