

# 1H-Cyclopropa[a]naphthalene, 1a,2,3,3a,4,5,6,7b-octahydro-1,1,3a,7-tetramethyl-, [1aR-(1a«alpha»,3a«alpha»,7b«alpha»)-1,1,3a,7-tetramethyl-]

Other names: «beta»-Maaliene

**Inchi:** InChI=1S/C15H24/c1-10-6-5-8-15(4)9-7-11-13(12(10)15)14(11,2)3/h11,13H,5-9H2,1-4H3

**InchiKey:** UPGLJTCDRBIZKP-UHFFFAOYSA-N

**Formula:** C15H24

**SMILES:** CC1=C2C3C(CCC2(C)CCC1)C3(C)C

**Mol. weight [g/mol]:** 204.35

**CAS:** 489-29-2

## Physical Properties

Property code	Value	Unit	Source
gf	225.48	kJ/mol	Joback Method
hf	-101.87	kJ/mol	Joback Method
hfus	13.73	kJ/mol	Joback Method
hvap	48.07	kJ/mol	Joback Method
log10ws	-4.67		Crippen Method
logp	4.559		Crippen Method
mcvol	185.330	ml/mol	McGowan Method
pc	2153.30	kPa	Joback Method
rinpol	1418.00		NIST Webbook
rinpol	1375.00		NIST Webbook
rinpol	1369.00		NIST Webbook
rinpol	1411.00		NIST Webbook
rinpol	1381.00		NIST Webbook
rinpol	1413.70		NIST Webbook
rinpol	1413.00		NIST Webbook
rinpol	1411.00		NIST Webbook
rinpol	1422.00		NIST Webbook
rinpol	1380.00		NIST Webbook
rinpol	1416.00		NIST Webbook
rinpol	1372.00		NIST Webbook
rinpol	1413.00		NIST Webbook
rinpol	1398.00		NIST Webbook
rinpol	1417.00		NIST Webbook
ripol	1671.00		NIST Webbook
tb	576.29	K	Joback Method

tc	802.89	K	Joback Method
tf	374.95	K	Joback Method
vc	0.712	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	503.46	J/mol×K	576.29	Joback Method
cpg	525.54	J/mol×K	614.06	Joback Method
cpg	546.15	J/mol×K	651.82	Joback Method
cpg	565.60	J/mol×K	689.59	Joback Method
cpg	584.18	J/mol×K	727.36	Joback Method
cpg	602.19	J/mol×K	765.12	Joback Method
cpg	619.93	J/mol×K	802.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C489292&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C489292&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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