

# Ethanethioamide, N,N-dimethyl-

<b>Other names:</b>	Acetamide, N,N-dimethylthio- Dimethylthioacetamide N,N-Dimethylthioacetamide Dimethylthioacetamid N,N-dimethylethanethioamide
<b>Inchi:</b>	InChI=1S/C4H9NS/c1-4(6)5(2)3/h1-3H3
<b>InchiKey:</b>	LKNQXZAHNDFIQY-UHFFFAOYSA-N
<b>Formula:</b>	C4H9NS
<b>SMILES:</b>	CC(=S)N(C)C
<b>Mol. weight [g/mol]:</b>	103.19
<b>CAS:</b>	631-67-4

## Physical Properties

Property code	Value	Unit	Source
affp	925.30	kJ/mol	NIST Webbook
basg	894.40	kJ/mol	NIST Webbook
gf	210.64	kJ/mol	Joback Method
hf	88.14	kJ/mol	Joback Method
hfus	13.74	kJ/mol	Joback Method
hvap	33.27	kJ/mol	Joback Method
ie	7.86	eV	NIST Webbook
log10ws	-0.92		Crippen Method
logp	0.895		Crippen Method
mcvol	89.250	ml/mol	McGowan Method
pc	4374.18	kPa	Joback Method
tb	373.40	K	Joback Method
tc	568.30	K	Joback Method
tf	201.58	K	Joback Method
vc	0.314	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	147.86	J/mol×K	373.40	Joback Method

cpg	157.40	J/mol×K	405.88	Joback Method
cpg	166.32	J/mol×K	438.37	Joback Method
cpg	174.64	J/mol×K	470.85	Joback Method
cpg	182.41	J/mol×K	503.33	Joback Method
cpg	189.65	J/mol×K	535.82	Joback Method
cpg	196.42	J/mol×K	568.30	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	399.00 ± 1.00	K	2.90	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C631674&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C631674&amp;Units=SI</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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