

Fumaric acid, monoamide, N-(2-bromophenyl)-, 2,4,6-trichlorophenyl

Inchi:
ester

InChI=1S/C16H9BrCl3NO3/c17-10-3-1-2-4-13(10)21-14(22)5-6-15(23)24-16-11(19)7-9(1

InchiKey:

KPDQBWJRVINLGB-AATRIKPKSA-N

Formula:

C16H9BrCl3NO3

SMILES:

O=C(C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl)Nc1ccccc1Br

Mol. weight [g/mol]:

449.51

Physical Properties

Property code	Value	Unit	Source
gf	55.44	kJ/mol	Joback Method
hf	-153.97	kJ/mol	Joback Method
hfus	51.29	kJ/mol	Joback Method
hvap	100.30	kJ/mol	Joback Method
log10ws	-6.71		Crippen Method
logp	5.510		Crippen Method
mcvol	257.690	ml/mol	McGowan Method
pc	2450.74	kPa	Joback Method
rinqol	3240.00		NIST Webbook
tb	1001.70	K	Joback Method
tc	1263.33	K	Joback Method
tf	692.23	K	Joback Method
vc	0.970	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	632.60	J/molxK	1001.70	Joback Method
cpg	639.95	J/molxK	1045.30	Joback Method
cpg	646.53	J/molxK	1088.91	Joback Method
cpg	652.42	J/molxK	1132.51	Joback Method
cpg	657.72	J/molxK	1176.12	Joback Method
cpg	662.51	J/molxK	1219.72	Joback Method
cpg	666.87	J/molxK	1263.33	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357467&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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