

# Pentane, 1-[(1-methylethyl)thio]-

<b>Other names:</b>	1-(Isopropylsulfanyl)pentane 2-Methyl-3-thiooctane Pentyl isopropyl sulfide Sulfide, isopropyl pentyl
<b>Inchi:</b>	InChI=1S/C8H18S/c1-4-5-6-7-9-8(2)3/h8H,4-7H2,1-3H3
<b>InchiKey:</b>	RLUYWKZJDJALNT-UHFFFAOYSA-N
<b>Formula:</b>	C8H18S
<b>SMILES:</b>	CCCCCSC(C)C
<b>Mol. weight [g/mol]:</b>	146.29
<b>CAS:</b>	7352-00-3

## Physical Properties

Property code	Value	Unit	Source
gf	47.16	kJ/mol	Joback Method
hf	-171.86	kJ/mol	Joback Method
hfus	17.08	kJ/mol	Joback Method
hvap	39.83	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	3.318		Crippen Method
mcvol	139.930	ml/mol	McGowan Method
pc	2600.43	kPa	Joback Method
rinpol	1034.00		NIST Webbook
rinpol	1034.00		NIST Webbook
rinpol	1034.00		NIST Webbook
rinpol	1034.00		NIST Webbook
tb	450.78	K	Joback Method
tc	640.58	K	Joback Method
tf	199.32	K	Joback Method
vc	0.531	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	287.04	J/molxK	450.78	Joback Method

cpg	301.31	J/mol×K	482.41	Joback Method
cpg	315.00	J/mol×K	514.05	Joback Method
cpg	328.11	J/mol×K	545.68	Joback Method
cpg	340.66	J/mol×K	577.31	Joback Method
cpg	352.65	J/mol×K	608.94	Joback Method
cpg	364.09	J/mol×K	640.58	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50110e+01
Coeff. B	-3.97703e+03
Coeff. C	-6.64450e+01
Temperature range (K), min.	336.56
Temperature range (K), max.	476.47

## Sources

The Yaws Handbook of Vapor

Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7352003&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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