

Butane, 1-bromo-3-methyl-

Other names:	1-Bromo-3-methylbutane 3-Methylbutyl bromide 4-Bromo-2-methylbutane BrCH ₂ CH ₂ CH(CH ₃) ₂ ISO-AMYL BROMIDE Isoamyl bromide Isopentyl bromide UN 2341
Inchi:	InChI=1S/C5H11Br/c1-5(2)3-4-6/h5H,3-4H2,1-2H3
InchiKey:	YXZFFTJAHVMMLF-UHFFFAOYSA-N
Formula:	C ₅ H ₁₁ Br
SMILES:	CC(C)CCBr
Mol. weight [g/mol]:	151.04
CAS:	107-82-4

Physical Properties

Property code	Value	Unit	Source
gf	3.10	kJ/mol	Joback Method
hf	-125.48	kJ/mol	Joback Method
hfus	10.47	kJ/mol	Joback Method
hvap	32.77	kJ/mol	Joback Method
log10ws	-2.89		Aqueous Solubility Prediction Method
logp	2.427		Crippen Method
mcvol	98.810	ml/mol	McGowan Method
pc	3805.69	kPa	Joback Method
rinpol	853.00		NIST Webbook
rinpol	796.00		NIST Webbook
rinpol	867.00		NIST Webbook
rinpol	837.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	794.00		NIST Webbook
ripol	1002.00		NIST Webbook
ripol	991.00		NIST Webbook
ripol	1009.00		NIST Webbook
ripol	1002.00		NIST Webbook
tb	393.40 ± 5.00	K	NIST Webbook

tb	393.80 ± 0.40	K	NIST Webbook
tb	393.45 ± 0.30	K	NIST Webbook
tb	394.60 ± 0.50	K	NIST Webbook
tb	393.75 ± 0.50	K	NIST Webbook
tb	393.50 ± 0.50	K	NIST Webbook
tb	393.50 ± 0.50	K	NIST Webbook
tb	393.80 ± 0.40	K	NIST Webbook
tb	393.50 ± 0.50	K	NIST Webbook
tb	393.75 ± 0.40	K	NIST Webbook
tb	392.00 ± 4.00	K	NIST Webbook
tb	393.80 ± 0.50	K	NIST Webbook
tb	393.60	K	NIST Webbook
tb	393.50 ± 0.50	K	NIST Webbook
tc	568.83	K	Joback Method
tf	190.91	K	Joback Method
vc	0.371	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	205.34	J/mol×K	537.28	Joback Method
cpg	197.35	J/mol×K	505.73	Joback Method
cpg	212.94	J/mol×K	568.83	Joback Method
cpg	161.30	J/mol×K	379.52	Joback Method
cpg	170.96	J/mol×K	411.07	Joback Method
cpg	180.18	J/mol×K	442.62	Joback Method
cpg	188.97	J/mol×K	474.18	Joback Method
cpl	187.00	J/mol×K	298.00	NIST Webbook
dvisc	0.0003481	Paxs	379.52	Joback Method
dvisc	0.0006202	Paxs	316.65	Joback Method
dvisc	0.0004527	Paxs	348.08	Joback Method
dvisc	0.0061616	Paxs	190.91	Joback Method
dvisc	0.0027207	Paxs	222.34	Joback Method
dvisc	0.0014710	Paxs	253.78	Joback Method
dvisc	0.0009108	Paxs	285.21	Joback Method
hvapt	36.60	kJ/mol	394.60	NIST Webbook
hvapt	41.00	kJ/mol	323.00	NIST Webbook
hvapt	37.90	kJ/mol	371.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47040e+01
Coeff. B	-3.55550e+03
Coeff. C	-4.14710e+01
Temperature range (K), min.	288.10
Temperature range (K), max.	420.02

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C*\ln(T) + D*T^2$
Coeff. A	8.26440e+01
Coeff. B	-7.33684e+03
Coeff. C	-1.01195e+01
Coeff. D	6.99212e-06
Temperature range (K), min.	306.15
Temperature range (K), max.	436.15

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C107824&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
KDB Vapor Pressure Data:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1621
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/files/research/kdb/mol/mol1621.mol
Aqueous Solubility Prediction Method:	http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cp_g: Ideal gas heat capacity

cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpolar:	Non-polar retention indices
ripolar:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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