

# Hexadecanoic acid, propyl ester

<b>Other names:</b>	Propyl hexadecanoate Propyl palmitate
<b>Inchi:</b>	InChI=1S/C19H38O2/c1-3-5-6-7-8-9-10-11-12-13-14-15-16-17-19(20)21-18-4-2/h3-18H2
<b>InchiKey:</b>	BEKZXQKGTDVSKX-UHFFFAOYSA-N
<b>Formula:</b>	C19H38O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(=O)OCCC
<b>Mol. weight [g/mol]:</b>	298.50
<b>CAS:</b>	2239-78-3

## Physical Properties

Property code	Value	Unit	Source
gf	-124.82	kJ/mol	Joback Method
hf	-680.29	kJ/mol	Joback Method
hfus	47.75	kJ/mol	Joback Method
hvap	67.04	kJ/mol	Joback Method
log10ws	-6.64		Crippen Method
logp	6.421		Crippen Method
mcvol	286.010	ml/mol	McGowan Method
pc	1114.08	kPa	Joback Method
rinpol	2091.00		NIST Webbook
rinpol	2091.30		NIST Webbook
rinpol	2077.00		NIST Webbook
rinpol	2065.00		NIST Webbook
rinpol	2072.00		NIST Webbook
rinpol	2094.00		NIST Webbook
rinpol	2091.00		NIST Webbook
ripol	2335.00		NIST Webbook
ripol	2348.00		NIST Webbook
tb	710.41	K	Joback Method
tc	879.90	K	Joback Method
tf	376.05	K	Joback Method
vc	1.123	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	951.34	J/molxK	879.90	Joback Method
cpg	936.27	J/molxK	851.65	Joback Method
cpg	920.41	J/molxK	823.40	Joback Method
cpg	903.75	J/molxK	795.15	Joback Method
cpg	886.25	J/molxK	766.91	Joback Method
cpg	867.91	J/molxK	738.66	Joback Method
cpg	848.69	J/molxK	710.41	Joback Method
dvisc	0.0018384	Paxs	376.05	Joback Method
dvisc	0.0000838	Paxs	710.41	Joback Method
dvisc	0.0001126	Paxs	654.68	Joback Method
dvisc	0.0001599	Paxs	598.96	Joback Method
dvisc	0.0002440	Paxs	543.23	Joback Method
dvisc	0.0004101	Paxs	487.50	Joback Method
dvisc	0.0007881	Paxs	431.78	Joback Method
hvapt	74.50	kJ/mol	458.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.83212e+01
Coeff. B	-6.83183e+03
Coeff. C	-1.23282e+02
Temperature range (K), min.	502.12
Temperature range (K), max.	648.41

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2239783&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.cheméo.com/doc/models/crippen\\_log10ws](https://www.cheméo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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