

# 2,2-Dimethylvaleric acid

<b>Other names:</b>	2,2-Dimethylpentanoic acid Pentanoic acid, 2,2-dimethyl- Valeric acid, 2,2-dimethyl- «alpha», «alpha»-Dimethylvaleric acid Â«alphaÂ», Â«alphaÂ»-Dimethylvaleric acid
<b>Inchi:</b>	InChI=1S/C7H14O2/c1-4-5-7(2,3)6(8)9/h4-5H2,1-3H3,(H,8,9)
<b>InchiKey:</b>	ZRYCZAWRXHAAPZ-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O2
<b>SMILES:</b>	CCCC(C)(C)C(=O)O
<b>Mol. weight [g/mol]:</b>	130.18
<b>CAS:</b>	1185-39-3

## Physical Properties

Property code	Value	Unit	Source
gf	-254.84	kJ/mol	Joback Method
hf	-461.37	kJ/mol	Joback Method
h <sub>fus</sub>	12.16	kJ/mol	Joback Method
h <sub>vap</sub>	53.31	kJ/mol	Joback Method
log <sub>10</sub> ws	-1.61		Crippen Method
logp	1.897		Crippen Method
m <sub>cvol</sub>	116.930	ml/mol	McGowan Method
pc	3395.98	kPa	Joback Method
tb	470.00	K	NIST Webbook
tb	420.60 ± 2.00	K	NIST Webbook
tb	423.70 ± 2.00	K	NIST Webbook
tc	682.23	K	Joback Method
tf	281.82	K	Joback Method
vc	0.442	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	268.88	J/mol×K	502.38	Joback Method
c <sub>pg</sub>	279.60	J/mol×K	532.36	Joback Method

cpg	289.76	J/mol×K	562.33	Joback Method
cpg	299.40	J/mol×K	592.31	Joback Method
cpg	308.52	J/mol×K	622.28	Joback Method
cpg	317.16	J/mol×K	652.26	Joback Method
cpg	325.33	J/mol×K	682.23	Joback Method
dvisc	0.0242619	Paxs	281.82	Joback Method
dvisc	0.0066100	Paxs	318.58	Joback Method
dvisc	0.0023568	Paxs	355.34	Joback Method
dvisc	0.0010196	Paxs	392.10	Joback Method
dvisc	0.0005092	Paxs	428.86	Joback Method
dvisc	0.0002838	Paxs	465.62	Joback Method
dvisc	0.0001723	Paxs	502.38	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.62687e+01
Coeff. B	-4.62077e+03
Coeff. C	-7.33810e+01
Temperature range (K), min.	362.52
Temperature range (K), max.	495.09

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1185393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1185393&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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