

# Bicyclo[3.1.0]hexan-2-one, 5-(1-methylethyl)-

<b>Other names:</b>	Bicyclo[3.1.0]hexan-2-one, 5-isopropyl-Sabina ketone 5-Isopropylbicyclo[3.1.0]hexan-2-one Sabine ketone
<b>Inchi:</b>	InChI=1S/C9H14O/c1-6(2)9-4-3-8(10)7(9)5-9/h6-7H,3-5H2,1-2H3
<b>InchiKey:</b>	MDDYCNAAAZKNAJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H14O
<b>SMILES:</b>	CC(C)C12CCC(=O)C1C2
<b>Mol. weight [g/mol]:</b>	138.21
<b>CAS:</b>	513-20-2

## Physical Properties

Property code	Value	Unit	Source
gf	15.88	kJ/mol	Joback Method
hf	-211.23	kJ/mol	Joback Method
hfus	5.02	kJ/mol	Joback Method
hvap	38.16	kJ/mol	Joback Method
log10ws	-1.93		Crippen Method
logp	2.012		Crippen Method
mcvol	117.520	ml/mol	McGowan Method
pc	3325.84	kPa	Joback Method
rinpol	1156.00		NIST Webbook
rinpol	1156.00		NIST Webbook
rinpol	1156.00		NIST Webbook
rinpol	1156.00		NIST Webbook
rinpol	1155.00		NIST Webbook
rinpol	1151.00		NIST Webbook
rinpol	1147.00		NIST Webbook
rinpol	1142.00		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	1126.00		NIST Webbook
rinpol	1160.00		NIST Webbook
rinpol	1165.00		NIST Webbook
rinpol	1179.00		NIST Webbook
rinpol	1154.00		NIST Webbook
rinpol	1160.00		NIST Webbook
rinpol	1156.00		NIST Webbook



ripol	1651.00		NIST Webbook
ripol	1651.00		NIST Webbook
ripol	1651.00		NIST Webbook
ripol	1651.00		NIST Webbook
ripol	1651.00		NIST Webbook
ripol	1651.00		NIST Webbook
tb	486.42	K	Joback Method
tc	710.14	K	Joback Method
tf	304.19	K	Joback Method
vc	0.453	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.21	J/mol×K	486.42	Joback Method
cpg	294.86	J/mol×K	523.71	Joback Method
cpg	310.28	J/mol×K	560.99	Joback Method
cpg	324.61	J/mol×K	598.28	Joback Method
cpg	338.00	J/mol×K	635.57	Joback Method
cpg	350.60	J/mol×K	672.85	Joback Method
cpg	362.55	J/mol×K	710.14	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C513202&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C513202&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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