

# Fumaric acid, monoamide, N,N-dimethyl-, 4-isopropoxyphenyl ester

Inchi:	InChI=1S/C15H19NO4/c1-11(2)19-12-5-7-13(8-6-12)20-15(18)10-9-14(17)16(3)4/h5-11H
InchiKey:	GSBUVGNPDVPIMO-MDZDMLPSA-N
Formula:	C15H19NO4
SMILES:	CC(C)Oc1ccc(OC(=O)C=CC(=O)N(C)C)cc1
Mol. weight [g/mol]:	277.32

## Physical Properties

Property code	Value	Unit	Source
gf	-101.08	kJ/mol	Joback Method
hf	-438.00	kJ/mol	Joback Method
hfus	33.53	kJ/mol	Joback Method
hvap	71.85	kJ/mol	Joback Method
log10ws	-2.73		Crippen Method
logp	2.024		Crippen Method
mcvol	219.010	ml/mol	McGowan Method
pc	2081.22	kPa	Joback Method
rinpol	2266.00		NIST Webbook
rinpol	2266.00		NIST Webbook
tb	743.00	K	Joback Method
tc	954.53	K	Joback Method
tf	454.46	K	Joback Method
vc	0.807	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	609.95	J/mol×K	743.00	Joback Method
cpg	624.56	J/mol×K	778.26	Joback Method
cpg	638.17	J/mol×K	813.51	Joback Method
cpg	650.79	J/mol×K	848.77	Joback Method
cpg	662.47	J/mol×K	884.02	Joback Method
cpg	673.24	J/mol×K	919.28	Joback Method
cpg	683.14	J/mol×K	954.53	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357420&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357420&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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