

# Phenol, 2-(1-methylethyl)-

<b>Other names:</b>	1-Hydroxy-2-isopropylbenzene 2-(1-Methylethyl)phenol 2-Isopropylphenol Isopropylphenol, ortho NSC 5103 Phenol, o-isopropyl- Prodox 131 o-Cumenol o-Hydroxycumene o-Isopropylphenol
<b>Inchi:</b>	InChI=1S/C9H12O/c1-7(2)8-5-3-4-6-9(8)10/h3-7,10H,1-2H3
<b>InchiKey:</b>	CRBJBYGJVIBWIY-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O
<b>SMILES:</b>	CC(C)c1ccccc1O
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	88-69-7

## Physical Properties

Property code	Value	Unit	Source
chl	-5022.90	kJ/mol	NIST Webbook
gf	-19.75	kJ/mol	Joback Method
hf	-175.30 ± 2.40	kJ/mol	NIST Webbook
hf	-161.30	kJ/mol	NIST Webbook
hfl	-230.00	kJ/mol	NIST Webbook
hfus	15.37	kJ/mol	Joback Method
hvap	68.70	kJ/mol	NIST Webbook
hvap	73.01	kJ/mol	NIST Webbook
log10ws	-2.20		Crippen Method
logp	2.516		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3935.71	kPa	Joback Method
rinpole	1175.00		NIST Webbook
rinpole	1200.00		NIST Webbook
rinpole	1175.00		NIST Webbook
rinpole	1198.00		NIST Webbook
rinpole	1200.00		NIST Webbook
rinpole	1173.00		NIST Webbook

rmpol	203.41		NIST Webbook
rmpol	1175.00		NIST Webbook
rmpol	1198.90		NIST Webbook
rmpol	1178.00		NIST Webbook
rmpol	1175.00		NIST Webbook
rmpol	1184.00		NIST Webbook
rmpol	1175.40		NIST Webbook
rmpol	1203.00		NIST Webbook
rmpol	1198.90		NIST Webbook
rmpol	1199.20		NIST Webbook
rmpol	1200.20		NIST Webbook
rmpol	1200.00		NIST Webbook
ripol	2089.00		NIST Webbook
ripol	2111.00		NIST Webbook
tb	512.18	K	Joback Method
tc	739.56	K	Joback Method
tf	288.65 ± 2.00	K	NIST Webbook
tf	288.65 ± 2.00	K	NIST Webbook
tf	288.65 ± 2.00	K	NIST Webbook
tf	290.00 ± 0.20	K	NIST Webbook
vc	0.392	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	325.34	J/molxK	701.66	Joback Method
cpg	293.23	J/molxK	587.97	Joback Method
cpg	280.87	J/molxK	550.08	Joback Method
cpg	267.53	J/molxK	512.18	Joback Method
cpg	304.71	J/molxK	625.87	Joback Method
cpg	334.65	J/molxK	739.56	Joback Method
cpg	315.38	J/molxK	663.76	Joback Method
dvisc	0.0072081	Paxs	314.33	Joback Method
dvisc	0.0000726	Paxs	512.18	Joback Method
dvisc	0.0001200	Paxs	479.21	Joback Method
dvisc	0.0002136	Paxs	446.23	Joback Method
dvisc	0.0004171	Paxs	413.25	Joback Method
dvisc	0.0009146	Paxs	380.28	Joback Method
dvisc	0.0023278	Paxs	347.31	Joback Method
hvapt	57.30	kJ/mol	418.00	NIST Webbook
hvapt	55.10	kJ/mol	429.50	NIST Webbook

hvapt	63.50	kJ/mol	434.00	NIST Webbook
hvapt	56.10	kJ/mol	434.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54296e+01
Coeff. B	-4.44272e+03
Coeff. C	-7.67170e+01
Temperature range (K), min.	370.12
Temperature range (K), max.	515.80

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C88697&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C88697&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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