

# Butanoic acid, 2-chloro, 1,1-dimethylethyl ester

Inchi:	InChI=1S/C8H15ClO2/c1-5-6(9)7(10)11-8(2,3)4/h6H,5H2,1-4H3
InchiKey:	VAUWEPUIEVLFBSB-UHFFFAOYSA-N
Formula:	C8H15ClO2
SMILES:	CCC(Cl)C(=O)OC(C)(C)C
Mol. weight [g/mol]:	178.66

## Physical Properties

Property code	Value	Unit	Source
gf	-228.97	kJ/mol	Joback Method
hf	-483.02	kJ/mol	Joback Method
hfus	12.52	kJ/mol	Joback Method
hvap	45.26	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	2.345		Crippen Method
mcvol	143.260	ml/mol	McGowan Method
pc	2629.85	kPa	Joback Method
rinpol	1012.00		NIST Webbook
rinpol	992.00		NIST Webbook
rinpol	997.00		NIST Webbook
rinpol	1008.00		NIST Webbook
rinpol	1003.00		NIST Webbook
ripol	1292.00		NIST Webbook
ripol	1287.00		NIST Webbook
ripol	1327.00		NIST Webbook
ripol	1299.00		NIST Webbook
tb	492.49	K	Joback Method
tc	688.63	K	Joback Method
tf	269.42	K	Joback Method
vc	0.539	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.00	J/molxK	492.49	Joback Method

cpg	324.20	J/mol×K	525.18	Joback Method
cpg	336.72	J/mol×K	557.87	Joback Method
cpg	348.58	J/mol×K	590.56	Joback Method
cpg	359.80	J/mol×K	623.25	Joback Method
cpg	370.40	J/mol×K	655.94	Joback Method
cpg	380.41	J/mol×K	688.63	Joback Method
dvisc	0.0057226	Paxs	269.42	Joback Method
dvisc	0.0024535	Paxs	306.60	Joback Method
dvisc	0.0012634	Paxs	343.78	Joback Method
dvisc	0.0007405	Paxs	380.96	Joback Method
dvisc	0.0004773	Paxs	418.13	Joback Method
dvisc	0.0003305	Paxs	455.31	Joback Method
dvisc	0.0002420	Paxs	492.49	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R28540&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R28540&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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