

# 1-Phenanthrenecarboxylic acid, 1,2,3,4,4a,4b,5,6,7,9,10,10a-dodecahydro-1,4a-dimethyl ester [1R-(1«alpha»,4a«beta»,4b«alpha»,10a«alpha»)]-

Other names:

Podocarp-8(14)-en-15-oic acid, 13-isopropylidene-, methyl ester

Methyl neoabietate

Podocarpic acid, methyl ester

Inchi: InChI=1S/C21H32O2/c1-14(2)15-7-9-17-16(13-15)8-10-18-20(17,3)11-6-12-21(18,4)19(2)

InchiKey: XLNYKQDSHLEWFW-QDDLWIDSA-N

Formula: C21H32O2

SMILES: COC(=O)C1(C)CCCC2(C)C3CCC(=C(C)C)C=C3CCC12

Mol. weight [g/mol]: 316.48

CAS: 3310-97-2

## Physical Properties

Property code	Value	Unit	Source
gf	52.32	kJ/mol	Joback Method
hf	-411.28	kJ/mol	Joback Method
hfus	25.16	kJ/mol	Joback Method
hvap	71.31	kJ/mol	Joback Method
log10ws	-5.90		Crippen Method
logp	5.439		Crippen Method
mcvol	273.010	ml/mol	McGowan Method
pc	1524.69	kPa	Joback Method
rinpol	2387.00		NIST Webbook
rinpol	2431.00		NIST Webbook
rinpol	2431.00		NIST Webbook
rinpol	2387.00		NIST Webbook
rinpol	2397.00		NIST Webbook
rinpol	2439.00		NIST Webbook
rinpol	2439.00		NIST Webbook
ripol	3055.00		NIST Webbook
ripol	3062.00		NIST Webbook
tb	804.21	K	Joback Method
tc	1038.48	K	Joback Method
tf	488.05	K	Joback Method
vc	1.032	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	887.02	J/mol×K	804.21	Joback Method
cpg	911.81	J/mol×K	843.25	Joback Method
cpg	936.26	J/mol×K	882.30	Joback Method
cpg	960.68	J/mol×K	921.34	Joback Method
cpg	985.40	J/mol×K	960.39	Joback Method
cpg	1010.72	J/mol×K	999.43	Joback Method
cpg	1036.95	J/mol×K	1038.48	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3310972&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3310972&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/30-607-4/1-Phenanthrenecarboxylic-acid-1-2-3-4-4a-4b-5-6-7-9-10-10a-dodecahydro-1>

Generated by Cheméo on 2023-02-08 02:54:56.448132087 +0000 UTC m=+41234.444248417.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.