

# 3-Octanone

<b>Other names:</b>	3-Oxooctane 3-octanonere Amyl ethyl ketone EAK Ethyl amyl ketone Ethyl n-amyl ketone Ethyl n-pentyl ketone Ethyl pentyl ketone NSC 60161 Octan-3-one n-Octanone-3
<b>Inchi:</b>	InChI=1S/C8H16O/c1-3-5-6-7-8(9)4-2/h3-7H2,1-2H3
<b>InchiKey:</b>	RHLVCLIPMVJYKS-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O
<b>SMILES:</b>	CCCCC(=O)CC
<b>Mol. weight [g/mol]:</b>	128.21
<b>CAS:</b>	106-68-3

## Physical Properties

Property code	Value	Unit	Source
chl	-5052.01	kJ/mol	NIST Webbook
gf	-112.44	kJ/mol	Joback Method
hf	-338.60 ± 3.30	kJ/mol	NIST Webbook
hfus	18.07	kJ/mol	Joback Method
hvap	44.40 ± 0.20	kJ/mol	NIST Webbook
log10ws	-2.45		Crippen Method
logp	2.546		Crippen Method
mcvol	125.150	ml/mol	McGowan Method
pc	2704.22	kPa	Joback Method
rhoc	257.71 ± 2.56	kg/m3	NIST Webbook
rinpola	986.00		NIST Webbook
rinpola	985.00		NIST Webbook
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tb	436.31	K	Joback Method
tc	627.70 ± 0.20	K	NIST Webbook
tf	229.85	K	Joback Method
vc	0.489	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.17	J/molxK	582.49	Joback Method
cpg	302.57	J/molxK	553.25	Joback Method
cpg	291.52	J/molxK	524.02	Joback Method
cpg	280.00	J/molxK	494.78	Joback Method
cpg	267.99	J/molxK	465.55	Joback Method
cpg	255.50	J/molxK	436.31	Joback Method
cpg	323.31	J/molxK	611.72	Joback Method
dvisc	0.0043730	Paxs	229.85	Joback Method
dvisc	0.0002948	Paxs	436.31	Joback Method
dvisc	0.0003813	Paxs	401.90	Joback Method
dvisc	0.0005173	Paxs	367.49	Joback Method
dvisc	0.0007477	Paxs	333.08	Joback Method
dvisc	0.0011762	Paxs	298.67	Joback Method
dvisc	0.0020821	Paxs	264.26	Joback Method
hvapt	43.80	kJ/mol	320.50	NIST Webbook
rho1	817.77	kg/m3	298.15	Excess Molar Enthalpies of Binary Systems of 2-Octanone or 3-Octanone with Dodecane, Tetradecane, or Hexadecane at 298.15 K

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52035e+01



Coeff. B	-3.97315e+03
Coeff. C	-6.36510e+01
Temperature range (K), min.	330.02
Temperature range (K), max.	465.30

## Sources

Excess Molar Enthalpies of Binary Systems of 2-Octanone or 3-Octanone with Decane, Tetradecane, or Hexadecane at 298.15 K: McGowan Method:	<a href="https://www.doi.org/10.1021/je900311v">https://www.doi.org/10.1021/je900311v</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
NIST Webbook:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
The Yaws Handbook of Vapor Pressure: Crippen Method:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C106683&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C106683&amp;Units=SI</a>
Crippen Method:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rhoc:</b>	Critical density
<b>rhol:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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