

# 2-Cyclohexen-1-ol, 2-methyl-5-(1-methylethenyl)-, acetate

<b>Other names:</b>	p-Mentha-6,8-dien-2-ol, acetate Carveol acetate Carvyl acetate L-Carvyl acetate 6-Acetoxy-p-mentha-1,8-diene p-Mentha-6,8-dien-2-yl acetate 6-Acetoxy-p-menta-1,8-diene p-mentha-1(6),8-dien-2-yl acetate
<b>Inchi:</b>	InChI=1S/C12H18O2/c1-8(2)11-6-5-9(3)12(7-11)14-10(4)13/h5,11-12H,1,6-7H2,2-4H3
<b>InchiKey:</b>	YTHRBOFHFYZBRJ-UHFFFAOYSA-N
<b>Formula:</b>	C12H18O2
<b>SMILES:</b>	<chem>C=C(C)C1CC=C(C)C(OC(C)=O)C1</chem>
<b>Mol. weight [g/mol]:</b>	194.27
<b>CAS:</b>	97-42-7

## Physical Properties

Property code	Value	Unit	Source
gf	-67.40	kJ/mol	Joback Method
hf	-339.88	kJ/mol	Joback Method
hfus	20.77	kJ/mol	Joback Method
hvap	51.95	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	2.850		Crippen Method
mcvol	167.920	ml/mol	McGowan Method
pc	2302.53	kPa	Joback Method
rinpol	1339.00		NIST Webbook
rinpol	1339.00		NIST Webbook
rinpol	1340.00		NIST Webbook
rinpol	1345.00		NIST Webbook
rinpol	1334.00		NIST Webbook
rinpol	1337.00		NIST Webbook
rinpol	1333.00		NIST Webbook
rinpol	1341.00		NIST Webbook
rinpol	1334.00		NIST Webbook
ripol	1368.00		NIST Webbook
ripol	1759.00		NIST Webbook
ripol	1777.00		NIST Webbook

ripol	1759.00		NIST Webbook
tb	565.83	K	Joback Method
tc	775.66	K	Joback Method
tf	297.86	K	Joback Method
vc	0.631	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	415.61	J/mol×K	565.83	Joback Method
cpg	433.54	J/mol×K	600.80	Joback Method
cpg	450.52	J/mol×K	635.77	Joback Method
cpg	466.58	J/mol×K	670.74	Joback Method
cpg	481.72	J/mol×K	705.71	Joback Method
cpg	495.96	J/mol×K	740.68	Joback Method
cpg	509.31	J/mol×K	775.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97427&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97427&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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