

# p-Chlorophenyl 2,3-epoxypropyl ether

<b>Other names:</b>	1,2-Epoxy-3-(p-chlorophenoxy)propane Oxirane, [(4-chlorophenoxy)methyl]- Propane, 1-(p-chlorophenoxy)-2,3-epoxy- 2-((4-Chlorophenoxy)methyl)oxirane Chlorophenyl glycidyl ether p-Chlorophenyl glycidyl ether 4-Chlorophenyl glycidyl ether 1,2-Epoxy-3-chlorophenoxypropane 1-(p-Chlorophenoxy)-2,3-epoxypropane 4-Chlorophenyl 2,3-epoxypropyl ether Glycidyl p-chlorophenyl ether NSC 60260
<b>Inchi:</b>	InChI=1S/C9H9ClO2/c10-7-1-3-8(4-2-7)11-5-9-6-12-9/h1-4,9H,5-6H2
<b>InchiKey:</b>	KSLSZOOZWRMSAP-UHFFFAOYSA-N
<b>Formula:</b>	C9H9ClO2
<b>SMILES:</b>	Clc1ccc(OCC2CO2)cc1
<b>Mol. weight [g/mol]:</b>	184.62
<b>CAS:</b>	2212-05-7

## Physical Properties

Property code	Value	Unit	Source
gf	-14.62	kJ/mol	Joback Method
hf	-211.19	kJ/mol	Joback Method
hfus	24.22	kJ/mol	Joback Method
hvap	49.78	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	2.118		Crippen Method
mcvol	127.030	ml/mol	McGowan Method
pc	3464.28	kPa	Joback Method
tb	530.52	K	Joback Method
tc	758.72	K	Joback Method
tf	326.79	K	Joback Method
vc	0.476	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.82	J/molxK	530.52	Joback Method
cpg	291.76	J/molxK	568.55	Joback Method
cpg	303.82	J/molxK	606.59	Joback Method
cpg	315.05	J/molxK	644.62	Joback Method
cpg	325.48	J/molxK	682.66	Joback Method
cpg	335.18	J/molxK	720.69	Joback Method
cpg	344.18	J/molxK	758.72	Joback Method
dvisc	0.0017216	Paxs	326.79	Joback Method
dvisc	0.0012381	Paxs	360.75	Joback Method
dvisc	0.0009424	Paxs	394.70	Joback Method
dvisc	0.0007490	Paxs	428.65	Joback Method
dvisc	0.0006157	Paxs	462.61	Joback Method
dvisc	0.0005199	Paxs	496.56	Joback Method
dvisc	0.0004486	Paxs	530.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2212057&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2212057&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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