

# Pentanoic acid, 2-hydroxy-4-methyl-, ethyl ester

<b>Other names:</b>	Ethyl 2-hydroxy-4-methylpentanoate ethyl 2-hydroxy-4-methylvalerate
<b>Inchi:</b>	InChI=1S/C8H16O3/c1-4-11-8(10)7(9)5-6(2)3/h6-7,9H,4-5H2,1-3H3
<b>InchiKey:</b>	QRHOWVDPHIXNEN-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O3
<b>SMILES:</b>	CCOC(=O)C(O)CC(C)C
<b>Mol. weight [g/mol]:</b>	160.21
<b>CAS:</b>	10348-47-7

## Physical Properties

Property code	Value	Unit	Source
gf	-359.14	kJ/mol	Joback Method
hf	-616.04	kJ/mol	Joback Method
hfus	16.30	kJ/mol	Joback Method
hvap	58.46	kJ/mol	Joback Method
log10ws	-1.17		Crippen Method
logp	0.956		Crippen Method
mcvol	136.890	ml/mol	McGowan Method
pc	2979.54	kPa	Joback Method
rinpol	1060.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1060.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1043.00		NIST Webbook
rinpol	1060.00		NIST Webbook
rinpol	1078.00		NIST Webbook
rinpol	1060.00		NIST Webbook
rinpol	1060.00		NIST Webbook
ripol	1547.00		NIST Webbook
ripol	1547.00		NIST Webbook
ripol	1547.00		NIST Webbook
ripol	1547.00		NIST Webbook
ripol	1515.00		NIST Webbook
ripol	1545.00		NIST Webbook
ripol	1562.00		NIST Webbook
ripol	1538.00		NIST Webbook
ripol	1547.00		NIST Webbook

tb	550.03	K	Joback Method
tc	724.80	K	Joback Method
tf	282.90	K	Joback Method
vc	0.514	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.04	J/mol×K	550.03	Joback Method
cpg	345.30	J/mol×K	579.16	Joback Method
cpg	356.10	J/mol×K	608.29	Joback Method
cpg	366.45	J/mol×K	637.42	Joback Method
cpg	376.36	J/mol×K	666.54	Joback Method
cpg	385.82	J/mol×K	695.67	Joback Method
cpg	394.85	J/mol×K	724.80	Joback Method
dvisc	0.0249265	Paxs	282.90	Joback Method
dvisc	0.0052844	Paxs	327.42	Joback Method
dvisc	0.0016241	Paxs	371.94	Joback Method
dvisc	0.0006423	Paxs	416.46	Joback Method
dvisc	0.0003039	Paxs	460.99	Joback Method
dvisc	0.0001640	Paxs	505.51	Joback Method
dvisc	0.0000978	Paxs	550.03	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10348477&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10348477&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/30-740-6/Pentanoic-acid-2-hydroxy-4-methyl-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-29 15:56:16.224511512 +0000 UTC m=+16695425.145088889.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.