

Sulfamide

Other names:	Sulfamamide Sulfonyl diamide Sulfuric diamide Sulfuryl amide Sulfuryl diamide sulphuric diamide
Inchi:	InChI=1S/H4N2O2S/c1-5(2,3)4/h(H4,1,2,3,4)
InchiKey:	NVBFHJWHLNUMCV-UHFFFAOYSA-N
Formula:	H4N2O2S
SMILES:	NS(N)(=O)=O
Mol. weight [g/mol]:	96.11
CAS:	7803-58-9

Physical Properties

Property code	Value	Unit	Source
gf	-386.52	kJ/mol	Joback Method
hf	-429.10	kJ/mol	Joback Method
hfus	17.53	kJ/mol	Joback Method
hvap	55.51	kJ/mol	Joback Method
log10ws	0.50		Crippen Method
logp	-1.851		Crippen Method
mcvol	58.910	ml/mol	McGowan Method
pc	11103.70	kPa	Joback Method
tb	392.24	K	Joback Method
tc	592.47	K	Joback Method
tf	294.84	K	Joback Method
vc	0.220	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	109.57	J/mol×K	392.24	Joback Method
cpg	114.15	J/mol×K	425.61	Joback Method
cpg	118.57	J/mol×K	458.98	Joback Method

cpg	122.85	J/mol×K	492.35	Joback Method
cpg	126.96	J/mol×K	525.73	Joback Method
cpg	130.89	J/mol×K	559.10	Joback Method
cpg	134.63	J/mol×K	592.47	Joback Method
hsubt	101.50 ± 1.00	kJ/mol	352.50	NIST Webbook

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C7803589&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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