

# Cyclopentane, (1,1-dimethylethyl)-

<b>Other names:</b>	tert-Butylcyclopentane
<b>Inchi:</b>	InChI=1S/C9H18/c1-9(2,3)8-6-4-5-7-8/h8H,4-7H2,1-3H3
<b>InchiKey:</b>	BFWVYBVSRYIDHI-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	CC(C)(C)C1CCCC1
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	3875-52-3

## Physical Properties

Property code	Value	Unit	Source
gf	64.29	kJ/mol	Joback Method
hf	-177.36	kJ/mol	Joback Method
hfus	5.59	kJ/mol	Joback Method
hvap	34.59	kJ/mol	Joback Method
log10ws	-3.00		Crippen Method
logp	3.223		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2820.33	kPa	Joback Method
rinpol	877.50		NIST Webbook
rinpol	881.30		NIST Webbook
tb	417.95 ± 0.20	K	NIST Webbook
tb	418.10 ± 1.00	K	NIST Webbook
tb	417.99 ± 0.20	K	NIST Webbook
tb	418.40 ± 1.00	K	NIST Webbook
tc	621.63	K	Joback Method
tf	204.51	K	Joback Method
vc	0.469	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.50	J/mol×K	417.37	Joback Method
cpg	343.64	J/mol×K	587.58	Joback Method
cpg	328.46	J/mol×K	553.54	Joback Method

cpg	312.30	J/molxK	519.50	Joback Method
cpg	295.12	J/molxK	485.46	Joback Method
cpg	276.87	J/molxK	451.41	Joback Method
cpg	357.89	J/molxK	621.63	Joback Method
dvisc	0.0003296	Paxs	417.37	Joback Method
dvisc	0.0004411	Paxs	381.89	Joback Method
dvisc	0.0006266	Paxs	346.42	Joback Method
dvisc	0.0009645	Paxs	310.94	Joback Method
dvisc	0.0016588	Paxs	275.46	Joback Method
dvisc	0.0033494	Paxs	239.99	Joback Method
dvisc	0.0086296	Paxs	204.51	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37426e+01
Coeff. B	-3.25078e+03
Coeff. C	-6.16710e+01
Temperature range (K), min.	303.28
Temperature range (K), max.	447.24

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol507.mol">https://www.thermo.com/files/research/kdb/mol/mol507.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3875523&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3875523&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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