

Butanoic acid, 3-hexenyl ester, (Z)-

Other names:	(3Z)-3-Hexenyl butyrate (Z)-3-Hexen-1-ol, butanoate (Z)-3-Hexen-1-yl butanoate (Z)-3-Hexenyl butyrate (Z)-3-hexen-1-yl butyrate (Z)-3-hexenyl butanoate (Z)-Hex-3-enyl butanoate (Z)-Hex-3-enyl butyrate 3(Z)-Hexenyl butanoate 3-Hexen-1-ol, butanoate, (Z)- Butanoic acid, (Z)-3-hexenyl ester Butyric acid, 3-hexenyl ester, (Z)- Hex-3(Z)-enyl butanoate Hexenyl butanoate (3Z) cis-3-Hexen-1-yl butyrate cis-3-Hexenyl butanoate cis-3-Hexenyl n-butyrate cis-3-hexenyl butyrate cis-Butyric acid, 3-hexenyl ester cis-Hex-3-enyl butanoate cis-Hex-3-enyl butyrate
Inchi:	InChI=1S/C10H18O2/c1-3-5-6-7-9-12-10(11)8-4-2/h5-6H,3-4,7-9H2,1-2H3/b6-5-
InchiKey:	ZCHOPXVYTWUHDS-WAYWQWQTSA-N
Formula:	C10H18O2
SMILES:	CCC=CCCOC(=O)CCC
Mol. weight [g/mol]:	170.25
CAS:	16491-36-4

Physical Properties

Property code	Value	Unit	Source
gf	-120.38	kJ/mol	Joback Method
hf	-377.31	kJ/mol	Joback Method
hfus	24.64	kJ/mol	Joback Method
hvap	46.97	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	2.686		Crippen Method

mcvol	154.900	ml/mol	McGowan Method
pc	2311.39	kPa	Joback Method
rinpol	1179.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1170.00		NIST Webbook
rinpol	1178.00		NIST Webbook
rinpol	1168.00		NIST Webbook
rinpol	1166.00		NIST Webbook
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ripol	1464.00		NIST Webbook
tb	508.65	K	Joback Method
tc	688.43	K	Joback Method
tf	269.54	K	Joback Method
vc	0.600	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.74	J/molxK	508.65	Joback Method
cpg	364.56	J/molxK	538.61	Joback Method
cpg	377.79	J/molxK	568.58	Joback Method
cpg	390.44	J/molxK	598.54	Joback Method
cpg	402.54	J/molxK	628.50	Joback Method
cpg	414.10	J/molxK	658.47	Joback Method
cpg	425.13	J/molxK	688.43	Joback Method
dvisc	0.0030444	Paxs	269.54	Joback Method
dvisc	0.0014196	Paxs	309.39	Joback Method
dvisc	0.0007878	Paxs	349.24	Joback Method
dvisc	0.0004933	Paxs	389.10	Joback Method
dvisc	0.0003369	Paxs	428.95	Joback Method
dvisc	0.0002455	Paxs	468.80	Joback Method

dvisc	0.0001880	Paxs	508.65	Joback Method
hvapt	59.90	kJ/mol	298.15	Vapor pressures and vaporization enthalpies of a series of esters used in flavors by correlation gas chromatography

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C16491364&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Vapor pressures and vaporization enthalpies of a series of esters used in flavors by correlation gas chromatography:	https://www.doi.org/10.1016/j.jct.2015.02.015 https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpolar:	Non-polar retention indices
ripolar:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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