

# 2,4-Pentanedione, 3-phenyl-

<b>Other names:</b>	3-Phenyl-2,4-pentanedione 3-Phenylacetylacetone
<b>Inchi:</b>	InChI=1S/C11H12O2/c1-8(12)11(9(2)13)10-6-4-3-5-7-10/h3-7,11H,1-2H3
<b>InchiKey:</b>	YIWTXSVNRCWBAC-UHFFFAOYSA-N
<b>Formula:</b>	C11H12O2
<b>SMILES:</b>	CC(=O)C(C(C)=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	176.21
<b>CAS:</b>	5910-25-8

## Physical Properties

Property code	Value	Unit	Source
gf	-106.13	kJ/mol	Joback Method
hf	-264.28	kJ/mol	Joback Method
hfus	17.96	kJ/mol	Joback Method
hvap	55.46	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	1.948		Crippen Method
mvol	145.230	ml/mol	McGowan Method
pc	3055.79	kPa	Joback Method
rinpol	1633.00		NIST Webbook
tb	585.06	K	Joback Method
tc	809.43	K	Joback Method
tf	325.01	K	Joback Method
vc	0.549	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	339.88	J/mol×K	585.06	Joback Method
cpg	353.65	J/mol×K	622.46	Joback Method
cpg	366.48	J/mol×K	659.85	Joback Method
cpg	378.41	J/mol×K	697.25	Joback Method
cpg	389.48	J/mol×K	734.64	Joback Method
cpg	399.74	J/mol×K	772.04	Joback Method

cpg	409.22	J/mol×K	809.43	Joback Method
dvisc	0.0033356	Paxs	325.01	Joback Method
dvisc	0.0016707	Paxs	368.35	Joback Method
dvisc	0.0009679	Paxs	411.69	Joback Method
dvisc	0.0006222	Paxs	455.03	Joback Method
dvisc	0.0004319	Paxs	498.38	Joback Method
dvisc	0.0003179	Paxs	541.72	Joback Method
dvisc	0.0002448	Paxs	585.06	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5910258&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5910258&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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