

# 1-Octene, 6-methyl-

<b>Other names:</b>	6-Methyl-1-octene
<b>Inchi:</b>	InChI=1S/C9H18/c1-4-6-7-8-9(3)5-2/h4,9H,1,5-8H2,2-3H3
<b>InchiKey:</b>	NFTYCFXCGQYTKV-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	C=CCCCC(C)CC
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	13151-10-5

## Physical Properties

Property code	Value	Unit	Source
gf	110.30	kJ/mol	Joback Method
hf	-108.94	kJ/mol	Joback Method
hfus	14.26	kJ/mol	Joback Method
hvap	34.57	kJ/mol	Joback Method
log10ws	-3.20		Crippen Method
logp	3.389		Crippen Method
mcvol	133.370	ml/mol	McGowan Method
pc	2421.88	kPa	Joback Method
rinpola	853.00		NIST Webbook
rinpola	853.00		NIST Webbook
rinpola	850.00		NIST Webbook
rinpola	853.00		NIST Webbook
rinpola	853.00		NIST Webbook
tb	401.56	K	Joback Method
tc	571.62	K	Joback Method
tf	174.43	K	Joback Method
vc	0.514	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.23	J/molxK	401.56	Joback Method
cpg	321.86	J/molxK	543.28	Joback Method
cpg	309.99	J/molxK	514.93	Joback Method

cpg	297.60	J/mol×K	486.59	Joback Method
cpg	284.69	J/mol×K	458.25	Joback Method
cpg	271.24	J/mol×K	429.90	Joback Method
cpg	333.24	J/mol×K	571.62	Joback Method
dvisc	0.0002302	Paxs	401.56	Joback Method
dvisc	0.0003086	Paxs	363.70	Joback Method
dvisc	0.0004428	Paxs	325.85	Joback Method
dvisc	0.0006987	Paxs	288.00	Joback Method
dvisc	0.0012657	Paxs	250.14	Joback Method
dvisc	0.0028336	Paxs	212.28	Joback Method
dvisc	0.0090007	Paxs	174.43	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.59744e+01
Coeff. B	-4.07439e+03
Coeff. C	-5.87860e+01
Temperature range (K), min.	318.52
Temperature range (K), max.	440.90

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151105&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151105&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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