

# Cyclohexane, methylene-

<b>Other names:</b>	1-Methylenecyclohexane Methylenecyclohexane
<b>Inchi:</b>	InChI=1S/C7H12/c1-7-5-3-2-4-6-7/h1-6H2
<b>InchiKey:</b>	YULMNMJFAZWLLN-UHFFFAOYSA-N
<b>Formula:</b>	C7H12
<b>SMILES:</b>	C=C1CCCCC1
<b>Mol. weight [g/mol]:</b>	96.17
<b>CAS:</b>	1192-37-6

## Physical Properties

Property code	Value	Unit	Source
chl	-4412.40	kJ/mol	NIST Webbook
gf	93.30	kJ/mol	Joback Method
hf	-28.91	kJ/mol	Joback Method
hfus	3.49	kJ/mol	Joback Method
hvap	36.10 ± 0.30	kJ/mol	NIST Webbook
hvap	36.10 ± 0.30	kJ/mol	NIST Webbook
ie	9.18	eV	NIST Webbook
ie	9.13	eV	NIST Webbook
ie	9.12 ± 0.02	eV	NIST Webbook
ie	9.08 ± 0.01	eV	NIST Webbook
ie	8.80	eV	NIST Webbook
ie	9.13	eV	NIST Webbook
ie	9.04 ± 0.03	eV	NIST Webbook
ie	8.97 ± 0.01	eV	NIST Webbook
ie	8.94	eV	NIST Webbook
ie	9.70	eV	NIST Webbook
ie	8.93 ± 0.01	eV	NIST Webbook
log10ws	-2.50		Crippen Method
logp	2.507		Crippen Method
mcvol	94.330	ml/mol	McGowan Method
pc	3749.97	kPa	Joback Method
rinpol	736.00		NIST Webbook
rinpol	736.00		NIST Webbook
rinpol	741.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	741.70		NIST Webbook

rinpol	731.00		NIST Webbook
rinpol	750.00		NIST Webbook
rinpol	726.00		NIST Webbook
rinpol	745.00		NIST Webbook
rinpol	739.00		NIST Webbook
rinpol	745.00		NIST Webbook
rinpol	738.00		NIST Webbook
rinpol	750.00		NIST Webbook
rinpol	737.50		NIST Webbook
rinpol	745.00		NIST Webbook
rinpol	758.00		NIST Webbook
rinpol	750.80		NIST Webbook
rinpol	744.00		NIST Webbook
rinpol	750.00		NIST Webbook
rinpol	745.00		NIST Webbook
rinpol	750.00		NIST Webbook
rinpol	754.10		NIST Webbook
rinpol	759.30		NIST Webbook
rinpol	737.80		NIST Webbook
rinpol	743.00		NIST Webbook
rinpol	738.00		NIST Webbook
rinpol	750.00		NIST Webbook
rinpol	744.00		NIST Webbook
rinpol	745.50		NIST Webbook
tb	376.00 ± 3.00	K	NIST Webbook
tb	376.11 ± 0.20	K	NIST Webbook
tb	378.15 ± 6.00	K	NIST Webbook
tb	375.70 ± 0.70	K	NIST Webbook
tb	376.00 ± 3.00	K	NIST Webbook
tb	375.40 ± 4.00	K	NIST Webbook
tb	376.00 ± 4.00	K	NIST Webbook
tb	382.00 ± 5.00	K	NIST Webbook
tb	371.00 ± 1.00	K	NIST Webbook
tb	375.10 ± 2.00	K	NIST Webbook
tb	376.50 ± 1.00	K	NIST Webbook
tb	376.00 ± 3.00	K	NIST Webbook
tb	375.00 ± 3.00	K	NIST Webbook
tb	378.00 ± 6.00	K	NIST Webbook
tb	377.00 ± 5.00	K	NIST Webbook
tb	375.70	K	NIST Webbook
tb	379.00 ± 5.00	K	NIST Webbook
tb	379.00 ± 4.00	K	NIST Webbook
tb	376.17 ± 0.05	K	NIST Webbook
tb	379.00 ± 4.00	K	NIST Webbook

tc	588.02	K	Joback Method
tf	168.43 ± 0.03	K	NIST Webbook
tf	168.44 ± 0.02	K	NIST Webbook
tf	166.50 ± 0.50	K	NIST Webbook
vc	0.345	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	236.60	J/mol×K	588.02	Joback Method
cpg	225.63	J/mol×K	553.84	Joback Method
cpg	214.03	J/mol×K	519.66	Joback Method
cpg	201.79	J/mol×K	485.48	Joback Method
cpg	188.89	J/mol×K	451.30	Joback Method
cpg	175.31	J/mol×K	417.12	Joback Method
cpg	161.02	J/mol×K	382.94	Joback Method
cpl	177.40	J/mol×K	298.15	NIST Webbook
dvisc	0.0002894	Paxs	382.94	Joback Method
dvisc	0.0003823	Paxs	351.44	Joback Method
dvisc	0.0005334	Paxs	319.94	Joback Method
dvisc	0.0008004	Paxs	288.44	Joback Method
dvisc	0.0013267	Paxs	256.95	Joback Method
dvisc	0.0025327	Paxs	225.45	Joback Method
dvisc	0.0059647	Paxs	193.95	Joback Method
hvapt	34.40	kJ/mol	359.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37327e+01
Coeff. B	-2.91341e+03
Coeff. C	-5.60000e+01
Temperature range (K), min.	272.69
Temperature range (K), max.	401.96

# Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1192376&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1192376&amp;Units=SI</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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