

Methoxyacetamide, N,N-dihexyl-

Inchi:	InChI=1S/C15H31NO2/c1-4-6-8-10-12-16(15(17)14-18-3)13-11-9-7-5-2/h4-14H2,1-3H3
InchiKey:	FPALQRVJINGISK-UHFFFAOYSA-N
Formula:	C15H31NO2
SMILES:	CCCCCN(CCCCC)C(=O)COC
Mol. weight [g/mol]:	257.41

Physical Properties

Property code	Value	Unit	Source
gf	-47.72	kJ/mol	Joback Method
hf	-530.20	kJ/mol	Joback Method
hfus	40.41	kJ/mol	Joback Method
hvap	60.18	kJ/mol	Joback Method
log10ws	-3.53		Crippen Method
logp	3.622		Crippen Method
mvol	239.630	ml/mol	McGowan Method
pc	1463.49	kPa	Joback Method
rinpol	1837.00		NIST Webbook
rinpol	1837.00		NIST Webbook
tb	631.33	K	Joback Method
tc	798.13	K	Joback Method
tf	363.44	K	Joback Method
vc	0.917	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	660.66	J/mol×K	631.33	Joback Method
cpg	678.37	J/mol×K	659.13	Joback Method
cpg	695.32	J/mol×K	686.93	Joback Method
cpg	711.52	J/mol×K	714.73	Joback Method
cpg	726.99	J/mol×K	742.53	Joback Method
cpg	741.74	J/mol×K	770.33	Joback Method
cpg	755.80	J/mol×K	798.13	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308492&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cp_g:	Ideal gas heat capacity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
rin_{pol}:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/30-847-8/Methoxyacetamide-N-N-dihexyl.pdf>

Generated by Cheméo on 2025-12-25 04:57:14.695120388 +0000 UTC m=+6386832.225161042.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.