

# Propanoic acid, 3-chloro-2,2-dimethyl-

<b>Other names:</b>	Propionic acid, 3-chloro-2,2-dimethyl- «beta»-Chloro-«alpha», «alpha»-dimethylpropionic acid «beta»-Chloropivalic acid Chloropivalic acid Chlorotrimethylacetic acid 3-Chloro-2,2-dimethylpropionic acid 3-Chloropivalic acid NSC 89696 3-chloro-2,2-dimethylpropanoic acid
<b>Inchi:</b>	InChI=1S/C5H9ClO2/c1-5(2,3-6)4(7)8/h3H2,1-2H3,(H,7,8)
<b>InchiKey:</b>	YBJGQSNSAWZZHL-UHFFFAOYSA-N
<b>Formula:</b>	C5H9ClO2
<b>SMILES:</b>	CC(C)(CCl)C(=O)O
<b>Mol. weight [g/mol]:</b>	136.58
<b>CAS:</b>	13511-38-1

## Physical Properties

Property code	Value	Unit	Source
gf	-283.61	kJ/mol	Joback Method
hf	-435.83	kJ/mol	Joback Method
hfus	11.18	kJ/mol	Joback Method
hvap	53.24	kJ/mol	Joback Method
log10ws	-0.93		Crippen Method
logp	1.336		Crippen Method
mcvol	100.990	ml/mol	McGowan Method
pc	4103.88	kPa	Joback Method
tb	488.20	K	NIST Webbook
tc	683.23	K	Joback Method
tf	314.00 ± 4.00	K	NIST Webbook
vc	0.379	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	209.78	J/molxK	494.05	Joback Method
cpg	217.91	J/molxK	525.58	Joback Method
cpg	225.55	J/molxK	557.11	Joback Method
cpg	232.72	J/molxK	588.64	Joback Method
cpg	239.44	J/molxK	620.17	Joback Method
cpg	245.76	J/molxK	651.70	Joback Method
cpg	251.67	J/molxK	683.23	Joback Method
dvisc	0.0199748	Paxs	289.20	Joback Method
dvisc	0.0063037	Paxs	323.34	Joback Method
dvisc	0.0024796	Paxs	357.48	Joback Method
dvisc	0.0011477	Paxs	391.62	Joback Method
dvisc	0.0006011	Paxs	425.77	Joback Method
dvisc	0.0003465	Paxs	459.91	Joback Method
dvisc	0.0002156	Paxs	494.05	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	383.20	K	1.30	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13511381&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13511381&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>brp</sub>:</b>	Boiling point at reduced pressure
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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