

# Trichloroacetic acid, 4-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C8H4Cl4O2/c9-5-1-3-6(4-2-5)14-7(13)8(10,11)12/h1-4H
<b>InchiKey:</b>	UAMJIPWXLGMBPW-UHFFFAOYSA-N
<b>Formula:</b>	C8H4Cl4O2
<b>SMILES:</b>	O=C(Oc1ccc(Cl)cc1)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	273.93

## Physical Properties

Property code	Value	Unit	Source
gf	-159.54	kJ/mol	Joback Method
hf	-299.90	kJ/mol	Joback Method
hfus	22.29	kJ/mol	Joback Method
hvap	61.74	kJ/mol	Joback Method
log10ws	-4.03		Crippen Method
logp	3.616		Crippen Method
mcvol	156.220	ml/mol	McGowan Method
pc	3254.14	kPa	Joback Method
rinpol	1551.00		NIST Webbook
tb	636.88	K	Joback Method
tc	889.58	K	Joback Method
tf	413.12	K	Joback Method
vc	0.585	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	304.35	J/molxK	636.88	Joback Method
cpg	338.08	J/molxK	847.46	Joback Method
cpg	332.79	J/molxK	805.34	Joback Method
cpg	326.82	J/molxK	763.23	Joback Method
cpg	320.14	J/molxK	721.11	Joback Method
cpg	312.67	J/molxK	679.00	Joback Method
cpg	342.76	J/molxK	889.58	Joback Method
dvisc	0.0001871	Paxs	636.88	Joback Method
dvisc	0.0002351	Paxs	599.59	Joback Method

dvisc	0.0003046	Paxs	562.29	Joback Method
dvisc	0.0004093	Paxs	525.00	Joback Method
dvisc	0.0005755	Paxs	487.71	Joback Method
dvisc	0.0008563	Paxs	450.41	Joback Method
dvisc	0.0013686	Paxs	413.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307720&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307720&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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