

# Phthalic acid, 2,3-dimethylphenyl octyl ester

<b>Inchi:</b>	InChI=1S/C24H30O4/c1-4-5-6-7-8-11-17-27-23(25)20-14-9-10-15-21(20)24(26)28-22-16
<b>InchiKey:</b>	LKZUBSVJXTZQNY-UHFFFAOYSA-N
<b>Formula:</b>	C24H30O4
<b>SMILES:</b>	CCCCCCCCOC(=O)c1cccc1C(=O)Oc1cccc(C)c1C
<b>Mol. weight [g/mol]:</b>	382.49

## Physical Properties

Property code	Value	Unit	Source
gf	-120.71	kJ/mol	Joback Method
hf	-589.64	kJ/mol	Joback Method
hfus	50.40	kJ/mol	Joback Method
hvap	93.87	kJ/mol	Joback Method
log10ws	-7.68		Crippen Method
logp	6.040		Crippen Method
mcvol	316.380	ml/mol	McGowan Method
pc	1263.75	kPa	Joback Method
rinsol	2861.00		NIST Webbook
tb	969.40	K	Joback Method
tc	1193.86	K	Joback Method
tf	594.96	K	Joback Method
vc	1.212	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1014.67	J/molxK	969.40	Joback Method
cpg	1028.85	J/molxK	1006.81	Joback Method
cpg	1041.60	J/molxK	1044.22	Joback Method
cpg	1052.96	J/molxK	1081.63	Joback Method
cpg	1062.96	J/molxK	1119.04	Joback Method
cpg	1071.65	J/molxK	1156.45	Joback Method
cpg	1079.05	J/molxK	1193.86	Joback Method
dvisc	0.0002694	Paxs	594.96	Joback Method
dvisc	0.0001612	Paxs	657.37	Joback Method

dvisc	0.0001054	Paxs	719.77	Joback Method
dvisc	0.0000738	Paxs	782.18	Joback Method
dvisc	0.0000544	Paxs	844.59	Joback Method
dvisc	0.0000419	Paxs	906.99	Joback Method
dvisc	0.0000333	Paxs	969.40	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357090&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357090&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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