

Pentafluoropropanoic acid, 4-cyanophenyl ester

Inchi:	InChI=1S/C10H4F5NO2/c11-9(12,10(13,14)15)8(17)18-7-3-1-6(5-16)2-4-7/h1-4H
InchiKey:	QAKWNVVBWJTGSV-UHFFFAOYSA-N
Formula:	C10H4F5NO2
SMILES:	N#Cc1ccc(OC(=O)C(F)(F)C(F)(F)F)cc1
Mol. weight [g/mol]:	265.14

Physical Properties

Property code	Value	Unit	Source
gf	-933.01	kJ/mol	Joback Method
hf	-1102.64	kJ/mol	Joback Method
hfus	20.17	kJ/mol	Joback Method
hvap	53.75	kJ/mol	Joback Method
log10ws	-3.53		Crippen Method
logp	2.661		Crippen Method
mvol	145.670	ml/mol	McGowan Method
pc	2431.44	kPa	Joback Method
rinpol	938.00		NIST Webbook
tb	628.12	K	Joback Method
tc	831.50	K	Joback Method
tf	386.34	K	Joback Method
vc	0.606	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.03	J/mol×K	628.12	Joback Method
cpg	380.02	J/mol×K	662.02	Joback Method
cpg	388.25	J/mol×K	695.91	Joback Method
cpg	395.76	J/mol×K	729.81	Joback Method
cpg	402.61	J/mol×K	763.71	Joback Method
cpg	408.85	J/mol×K	797.61	Joback Method
cpg	414.51	J/mol×K	831.50	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308004&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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