

# Butane, 1,1,2,2,3,3,4,4-octafluoro-1,4-diiodo-

<b>Other names:</b>	1,4-Diiodooctafluorobutane 1,1,2,2,3,3,4,4-octafluoro-1,4-diiodobutane
<b>Inchi:</b>	InChI=1S/C4F8I2/c5-1(6,3(9,10)13)2(7,8)4(11,12)14
<b>InchiKey:</b>	JILAKKYYZPDQBE-UHFFFAOYSA-N
<b>Formula:</b>	C4F8I2
<b>SMILES:</b>	FC(F)(I)C(F)(F)C(F)(F)C(F)(F)I
<b>Mol. weight [g/mol]:</b>	453.84
<b>CAS:</b>	375-50-8

## Physical Properties

Property code	Value	Unit	Source
gf	-1448.08	kJ/mol	Joback Method
hf	-1576.03	kJ/mol	Joback Method
hfus	9.91	kJ/mol	Joback Method
hvap	31.52	kJ/mol	Joback Method
log10ws	-5.64		Crippen Method
logp	4.313		Crippen Method
mcvol	133.020	ml/mol	McGowan Method
pc	2684.64	kPa	Joback Method
rinpol	903.00		NIST Webbook
rinpol	904.00		NIST Webbook
rinpol	906.00		NIST Webbook
tb	458.44	K	Joback Method
tc	662.50	K	Joback Method
tf	265.36	K	Joback Method
vc	0.535	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.42	J/molxK	458.44	Joback Method
cpg	263.80	J/molxK	492.45	Joback Method
cpg	271.00	J/molxK	526.46	Joback Method
cpg	277.11	J/molxK	560.47	Joback Method

cpg	282.26	J/mol×K	594.48	Joback Method
cpg	286.53	J/mol×K	628.49	Joback Method
cpg	290.05	J/mol×K	662.50	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	358.00	K	13.30	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C375508&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C375508&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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