

# Decan-2-yl ethyl carbonate

<b>Inchi:</b>	InChI=1S/C13H26O3/c1-4-6-7-8-9-10-11-12(3)16-13(14)15-5-2/h12H,4-11H2,1-3H3
<b>InchiKey:</b>	FCVPQVAPQCTDDB-UHFFFAOYSA-N
<b>Formula:</b>	C13H26O3
<b>SMILES:</b>	CCCCCCCCC(C)OC(=O)OCC
<b>Mol. weight [g/mol]:</b>	230.34

## Physical Properties

Property code	Value	Unit	Source
gf	-282.78	kJ/mol	Joback Method
hf	-693.95	kJ/mol	Joback Method
hfus	29.88	kJ/mol	Joback Method
hvap	55.71	kJ/mol	Joback Method
log10ws	-4.30		Crippen Method
logp	4.299		Crippen Method
mcvol	207.340	ml/mol	McGowan Method
pc	1694.90	kPa	Joback Method
rinsol	1490.00		NIST Webbook
tb	595.11	K	Joback Method
tc	766.49	K	Joback Method
tf	315.66	K	Joback Method
vc	0.799	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.85	J/molxK	595.11	Joback Method
cpg	560.25	J/molxK	623.67	Joback Method
cpg	575.99	J/molxK	652.24	Joback Method
cpg	591.08	J/molxK	680.80	Joback Method
cpg	605.53	J/molxK	709.37	Joback Method
cpg	619.34	J/molxK	737.93	Joback Method
cpg	632.50	J/molxK	766.49	Joback Method
dvisc	0.0027073	Paxs	315.66	Joback Method
dvisc	0.0011685	Paxs	362.23	Joback Method

dvisc	0.0006108	Paxs	408.81	Joback Method
dvisc	0.0003646	Paxs	455.38	Joback Method
dvisc	0.0002395	Paxs	501.96	Joback Method
dvisc	0.0001689	Paxs	548.54	Joback Method
dvisc	0.0001259	Paxs	595.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373789&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373789&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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