

# Octane, 1-iodo-

<b>Other names:</b>	1-Iodooctane 1-Jodoktan 1-Octyl iodide 1-n-Octyl iodide Octyl iodide n-Octyl iodide
<b>Inchi:</b>	InChI=1S/C8H17I/c1-2-3-4-5-6-7-8-9/h2-8H2,1H3
<b>InchiKey:</b>	UWLHSHAHTBJTBA-UHFFFAOYSA-N
<b>Formula:</b>	C8H17I
<b>SMILES:</b>	CCCCCCCCI
<b>Mol. weight [g/mol]:</b>	240.12
<b>CAS:</b>	629-27-6

## Physical Properties

Property code	Value	Unit	Source
gf	74.60	kJ/mol	Joback Method
hf	-131.58	kJ/mol	Joback Method
hfus	20.88	kJ/mol	Joback Method
hvap	59.70	kJ/mol	NIST Webbook
log10ws	-4.12		Crippen Method
logp	3.782		Crippen Method
mcvol	149.400	ml/mol	McGowan Method
pc	2480.12	kPa	Joback Method
rinpol	1225.00		NIST Webbook
rinpol	1236.00		NIST Webbook
rinpol	1221.00		NIST Webbook
ripol	1493.00		NIST Webbook
ripol	1487.00		NIST Webbook
ripol	1508.00		NIST Webbook
tb	499.15 ± 0.60	K	NIST Webbook
tb	498.70	K	NIST Webbook
tc	671.37	K	Joback Method
tf	227.45 ± 0.50	K	NIST Webbook
vc	0.572	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	361.67	J/molxK	671.37	Joback Method
cpg	290.98	J/molxK	475.58	Joback Method
cpg	304.29	J/molxK	508.21	Joback Method
cpg	316.96	J/molxK	540.84	Joback Method
cpg	329.00	J/molxK	573.48	Joback Method
cpg	340.45	J/molxK	606.11	Joback Method
cpg	351.33	J/molxK	638.74	Joback Method
dvisc	0.0003119	Paxs	475.58	Joback Method
dvisc	0.0061362	Paxs	237.98	Joback Method
dvisc	0.0026208	Paxs	277.58	Joback Method
dvisc	0.0013843	Paxs	317.18	Joback Method
dvisc	0.0008425	Paxs	356.78	Joback Method
dvisc	0.0005662	Paxs	396.38	Joback Method
dvisc	0.0004090	Paxs	435.98	Joback Method
hvapt	50.70	kJ/mol	472.50	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	359.70	K	0.70	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47947e+01
Coeff. B	-4.26424e+03
Coeff. C	-7.96660e+01
Temperature range (K), min.	373.61
Temperature range (K), max.	529.33

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C629276&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C629276&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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