

# Benzenamine, N,N-diethyl-3-methyl-

<b>Other names:</b>	1-(Diethylamino)-3-methylbenzene 3-(Diethylamino)-1-methylbenzene 3-(Diethylamino)toluene 3-(N,N-Diethylamino)toluene 3-Methyl-N,N-diethylbenzenamine N,N-Diethyl-3-methylaniline N,N-Diethyl-m-toluidine NSC 96629 m-Methyl(diethylamino)benzene m-Methyl-N,N-diethylaniline m-Toluidine, N,N-diethyl-
<b>Inchi:</b>	InChI=1S/C11H17N/c1-4-12(5-2)11-8-6-7-10(3)9-11/h6-9H,4-5H2,1-3H3
<b>InchiKey:</b>	CIPVYROJHKLHJI-UHFFFAOYSA-N
<b>Formula:</b>	C11H17N
<b>SMILES:</b>	CCN(CC)c1cccc(C)c1
<b>Mol. weight [g/mol]:</b>	163.26
<b>CAS:</b>	91-67-8

## Physical Properties

Property code	Value	Unit	Source
affp	964.10	kJ/mol	NIST Webbook
basg	932.20	kJ/mol	NIST Webbook
gf	255.30	kJ/mol	Joback Method
hf	22.22	kJ/mol	Joback Method
hfus	20.92	kJ/mol	Joback Method
hvap	45.06	kJ/mol	Joback Method
ie	6.90	eV	NIST Webbook
log10ws	-2.69		Crippen Method
logp	2.841		Crippen Method
mcvol	152.070	ml/mol	McGowan Method
pc	2603.08	kPa	Joback Method
rinpola	1290.10		NIST Webbook
rinpola	1290.10		NIST Webbook
tb	495.18	K	Joback Method
tc	695.71	K	Joback Method
tf	285.14	K	Joback Method
vc	0.561	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.22	J/mol×K	495.18	Joback Method
cpg	350.65	J/mol×K	528.60	Joback Method
cpg	366.18	J/mol×K	562.02	Joback Method
cpg	380.85	J/mol×K	595.44	Joback Method
cpg	394.69	J/mol×K	628.86	Joback Method
cpg	407.74	J/mol×K	662.29	Joback Method
cpg	420.04	J/mol×K	695.71	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47193e+01
Coeff. B	-4.29023e+03
Coeff. C	-8.14430e+01
Temperature range (K), min.	378.72
Temperature range (K), max.	537.47

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C91678&Units=SI>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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